

Oxidation and Reduction Reactions: Gasoline Engines vs. Fuel Cells

(How do we recognize oxidation and reduction reactions?)

This activity requires that students know how to determine oxidation numbers.

Model 1: Examples of oxidation and reduction.

When a piece of zinc metal is placed in an aqueous solution of copper nitrate, solid copper appears deposited on the zinc metal and some of the zinc is found as $\text{Zn}^{2+}(\text{aq})$ in the solution.



Combustion of methane, a component of natural gas, with oxygen produces carbon dioxide, water and heat.



When hydrogen and oxygen gases are reacted in a hydrogen fuel cell, electrical energy is produced. The same reaction can occur when a spark initiates an explosion of hydrogen and oxygen. In either case the reaction of hydrogen and oxygen produce water.



Iron (III) oxide and aluminum react violently to produce liquid iron in this reaction.



Critical Thinking Questions

- Write the oxidation number above each atom's symbol in the four reactions (1) to (4).
- If an atom changes its oxidation number in the reaction, draw an arrow  from its symbol on the left side of the reaction to the corresponding symbol on the right side. Do this for all four reactions.
- Find and circle the reactant containing the atom whose oxidation number increases in each forward reaction.
- Find and put a square around the reactant containing the atom whose oxidation number decreases in each forward reaction.

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Organic Chemistry, a Guided Inquiry Andrei Straumanis, 2004 Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Chemistry** Richard S. Moog, John J. Farrell, 2014-01-13 Chemistry A Guided Approach 6th Edition follows the underlying principles developed by years of research on how readers learn and draws on testing by those using the POGIL methodology This text follows inquiry based learning and correspondingly emphasizes the underlying concepts and the reasoning behind the concepts This text offers an approach that follows modern cognitive learning principles by having readers learn how to create knowledge based on experimental data and how to test that knowledge *Introductory Chemistry* Michael P. Garoutte, Ashley B. Mahoney, 2015-08-10 The ChemActivities found in Introductory Chemistry A Guided Inquiry use the classroom guided inquiry approach and provide an excellent accompaniment to any one semester Introductory text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting General, Organic, and Biological Chemistry Michael P. Garoutte, 2014-02-24 Classroom activities to support a General Organic and Biological Chemistry text Students can follow a guided inquiry approach as they learn chemistry in the classroom General Organic and Biological Chemistry A Guided Inquiry serves as an accompaniment to a GOB Chemistry text It can suit the one or two semester course This supplemental text supports Process Oriented Guided Inquiry Learning POGIL which is a student focused group learning philosophy of instruction The materials offer ways to promote a student centered science classroom with activities The goal is for students to gain a greater understanding of chemistry through exploration *International Catalogue of Scientific Literature* , 1902 **Current Chemical Papers** Chemical Society (Great Britain), 1968 A classified world list of new papers in pure chemistry *Journal of Scientific & Industrial Research* , 1983 *Chemical Abstracts* , 1918 **Physical Chemistry, a Guided Inquiry** Spencer, James Nelson Spencer, 2003-01-22 **Journal of the Society of Chemical Industry** Society of Chemical Industry (Great Britain), 1920 Qualitative Chemical Analysis from the Standpoint of Solubilities, Ionization and Mass Action John Iredelle Dillard Hinds, 1913 *Prentice Hall Chemistry* , 2000 *Sustainable Technologies for Textile Wastewater Treatments* Subramanian Senthilkannan Muthu, 2021-03-23 Sustainable Technologies for Textile Wastewater Treatments takes on this complex and environmentally crucial issue by providing comprehensive coverage on new technologies and practices Sections provide technical detail and instruction on cutting edge technologies including innovative industrial uses of nanotechnology and waste biomass In addition case studies are provided on different textile wastewater treatment plants hence showing their full practical context Specific areas of discussion include zero liquid discharge nanomaterials adsorption and advanced oxidization processes AOP Appropriate case studies from textile wastewater treatment plants are included to help illustrate key points Other sections cover the cost of these methods before highlighting effective low cost options This book will be of

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1. Understanding the eBook Chemactivity 12 Solutions
 - The Rise of Digital Reading Chemactivity 12 Solutions
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemactivity 12 Solutions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemactivity 12 Solutions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemactivity 12 Solutions
 - Personalized Recommendations
 - Chemactivity 12 Solutions User Reviews and Ratings
 - Chemactivity 12 Solutions and Bestseller Lists
5. Accessing Chemactivity 12 Solutions Free and Paid eBooks

- Chemactivity 12 Solutions Public Domain eBooks
- Chemactivity 12 Solutions eBook Subscription Services
- Chemactivity 12 Solutions Budget-Friendly Options
- 6. Navigating Chemactivity 12 Solutions eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemactivity 12 Solutions Compatibility with Devices
 - Chemactivity 12 Solutions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemactivity 12 Solutions
 - Highlighting and Note-Taking Chemactivity 12 Solutions
 - Interactive Elements Chemactivity 12 Solutions
- 8. Staying Engaged with Chemactivity 12 Solutions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemactivity 12 Solutions
- 9. Balancing eBooks and Physical Books Chemactivity 12 Solutions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemactivity 12 Solutions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemactivity 12 Solutions
 - Setting Reading Goals Chemactivity 12 Solutions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemactivity 12 Solutions
 - Fact-Checking eBook Content of Chemactivity 12 Solutions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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