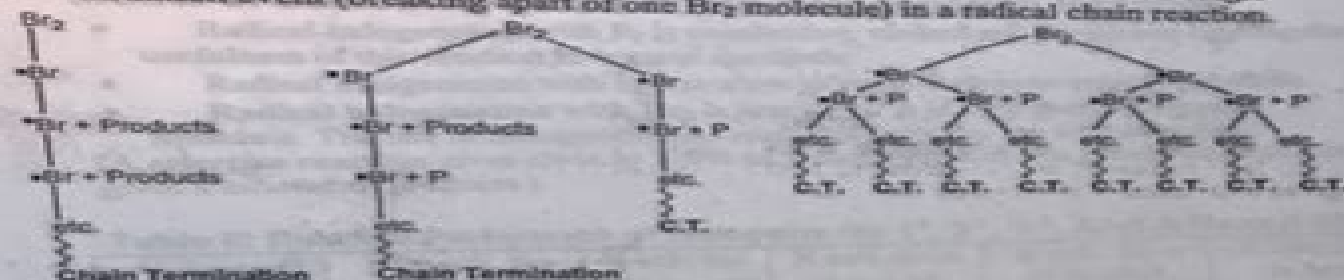


11. Which of the following diagrams best tracks the products formed from a single initiation event (breaking apart of one Br_2 molecule) in a radical chain reaction.



Model 7: Which H Will Be Replaced by X?

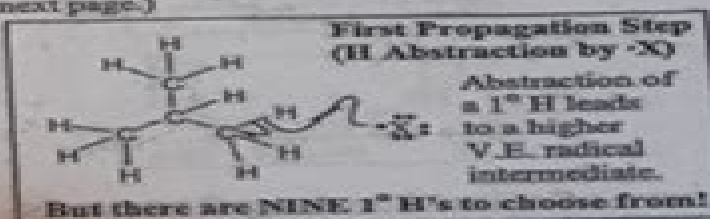
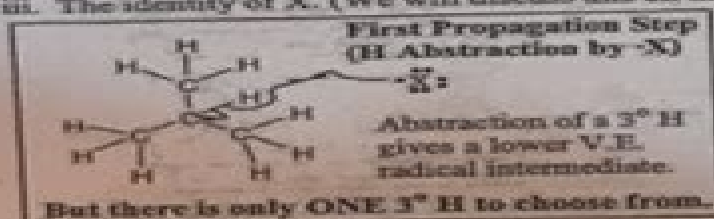
- In a radical halogenation reaction a halogen (F, Cl, Br or I) replaces an H.
- In most cases there is more than one type of H to choose from.

For example the hydrocarbon below has two different types of H's:



Three factors determine which H on a hydrocarbon will be replaced:

- The type of H. (benzyl, allyl, 3° , 2° , 1° or methyl)
- The number of H's of a given type.
- The identity of X. (We will discuss this on the next page.)



Critical Thinking Questions

12. Explain why, for the example above, you would expect a 1:9 ratio of 3° to 1° H substitution products if a 3° radical had about the same potential energy as a 1° radical.
13. But a 3° radical is LOWER in potential energy than a 1° radical. When $\text{X} = \text{Cl}$, a 3° radical intermediate is 5 times more favorable than a 1° radical intermediate. Given this, explain why, for the example above, a 5:9 ratio of 3° to 1° H substitution products is expected. (Consider both the type of H and the number of identical H's.)

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