

## Molarity

(How Concentrated Is It?)

Water is the most common solvent, and we will focus on aqueous solutions. However, the terms applied herein also apply when other solvents are used.

Solutes are categorized according to their ability to affect the electrical conductivity of the solution upon dissolution of the solute in water.

- An aqueous solution of a **strong electrolyte** conducts electricity well.
- An aqueous solution of a **nonelectrolyte** does not conduct electricity.
- An aqueous solution of a **weak electrolyte** conducts electricity poorly.
- Water from most common sources (tap water, rain water, water in lakes and rivers, etc.) always contains strong electrolytes and conducts electricity well.

### Model 1: Strong Electrolytes.

Many ionic compounds are strong electrolytes. When an ionic compound dissolves in water the cations and anions are separated. In the solution, the cations and anions are surrounded by water molecules. Sodium sulfate and sodium chloride are both strong electrolytes.



The "(aq)" after the ion indicates that each ion of that type is surrounded by several water molecules. It has been discovered that it is the mobile, charged particles in the solution that carry the electric current.

# Chemactivity 32 Molarity

**Richard Samuel Moog**



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*Chemistry* Richard S. Moog, John J. Farrell, 2014-01-13 *Chemistry A Guided Approach* 6th Edition follows the underlying principles developed by years of research on how readers learn and draws on testing by those using the POGIL methodology This text follows inquiry based learning and correspondingly emphasizes the underlying concepts and the reasoning behind the concepts This text offers an approach that follows modern cognitive learning principles by having readers learn how to create knowledge based on experimental data and how to test that knowledge *Merrill Chemistry* Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **General, Organic, and Biological Chemistry** Michael P. Garoutte, 2014-02-24 Classroom activities to support a General Organic and Biological Chemistry text Students can follow a guided inquiry approach as they learn chemistry in the classroom General Organic and Biological Chemistry A Guided Inquiry serves as an accompaniment to a GOB Chemistry text It can suit the one or two semester course This supplemental text supports Process Oriented Guided Inquiry Learning POGIL which is a student focused group learning philosophy of instruction The materials offer ways to promote a student centered science classroom with activities The goal is for students to gain a greater understanding of chemistry through exploration Prentice Hall Chemistry ,2000 Physical Chemistry, a Guided Inquiry Spencer, James Nelson Spencer, 2003-01-22 Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes

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