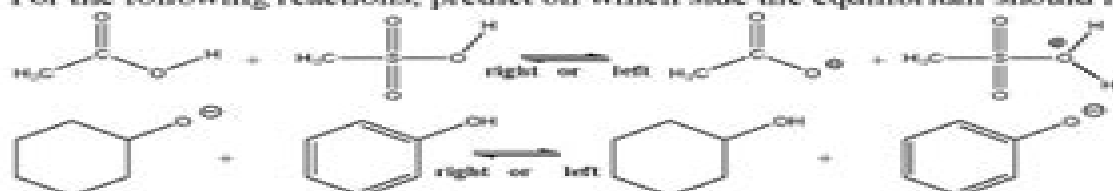


PS#2 ChemActivity 5-6

1. Rank the following set of molecules in order of acidity (1—most acidic).



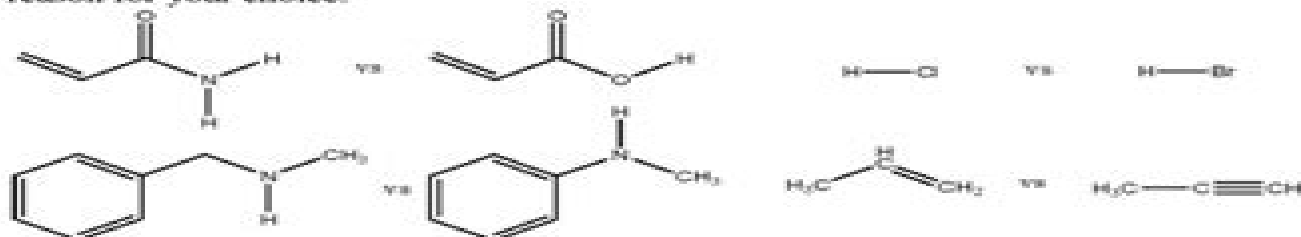
2. For the following reactions, predict on which side the equilibrium should lie. Explain.



3. Which is the stronger acid, A or B? Circle one.



- Draw ALL important resonance contributors of the conjugate base anions.
 - Draw curved arrows on the first structures to show the movement of electrons to obtain each of the following contributor.
 - Circle the most stable contributor for each.
 - Determine which is the stronger acid, A or B? Circle one and justify your answer.
4. For each pair, determine which is more acidic, circle the most acidic H and give a brief reason for your choice.



Chemactivity 5 Answers

**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**



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