#### Mole Calculation Worksheet

- 1) How many moles are in 15 grams of lithium?

  15 Light 1 mol L;
- 2) How many grams are in 2.4 moles of sulfur? 76.8 gS
- 3) How many moles are in 22 grams of argon?

  22 gAr ( i mol Ar ) = 0.55 mol Ar
- 4) How many grams are in 88.1 moles of magnesium?

  88.1 mol Mg ( 24 gMg ) = 2,114.4 gMg
- How many moles are in 2.3 grams of phosphorus?  $2.3gP\left(\frac{1 \text{ mol P}}{31 \text{ gP}}\right) = 0.1 \text{ mol P}$
- 6) How many grams are in 11.9 moles of chromium?

  11.9 mol ( 52 gCr )= 68.8 gCr
- 7) How many moles are in 9.8 grams of calcium?  $9.8gCa\left(\frac{1 \text{ mol Ca}}{52gCa}\right) = 0.2 \text{ mol Ca}$
- 8) How many grams are in 238 moles of arsenic?  $238 \text{ molAs} \left(\frac{75 \text{ gHs}}{1 \text{ suffs}}\right) = 17,850 \text{ g As}$

What are the molecular weights of the following compounds?

- 9) NaOH 40 grams 12) HoPO4 98 grams
- 10) H2O 18 grams 13) Mn2Ser 663 grams
- 11) MgCl2 95 grams 14) (NH4)2504 132 grams

# **Chemfiesta Mole Calculations Practice Answers**

**Richard Lloyd Whiteley** 

#### **Chemfiesta Mole Calculations Practice Answers:**

**Counting Moles** Nigel P. Freestone, 2016-01-04 Students studying chemistry often struggle with the mole Counting Moles provides an effective aid to learning by giving clear and confident presentation of the essentials of the mole concept needed by those starting chemistry courses This user friendly self teach e book is split into six chapters which sequentially introduce the mole calculating frame to help solve problems Over 200 fully worked examples are given along with several hundred questions The mole concept is applied to topics such as relative atomic mass and relative formula mass percentage composition empirical and molecular formula The book also covers concentration its units volumetric analysis and the relationship between volume mass and moles of gases Counting Moles culminates in you taking a Mole Driving Test On passing this test you are issued with a Counting Moles Driving License that will give you all the confidence required to correctly answer all mole calculations **Bite-Sized Chemistry Calculations** John Lambert, 2024-07-10 Bite Sized Chemistry Calculations is a series of books on chemistry calculations aimed at helping students overcome the challenges associated with tackling the various types of calculations encountered in different aspects of chemistry focusing on a few topics at a time to facilitate comprehension Written by an experienced chemistry educator each book in the series has been tailored to fully meet the needs of students at all levels especially those taking college level general chemistry courses as well as those following various O level curricula worldwide This part of the series explores the different types of problems and calculations encountered in mass the mole and stoichiometry including the determination of formulae of ionic compounds relative formula masses mass and percent compositions of compounds all aspects of mole calculations empirical and molecular formulae calculations based on chemical equations limiting reagents gas stoichiometry and percent yield The series is packed with many salient features that are meant to facilitate both teaching and learning Some of these include helpful explanations many examples alternative ways to solve problems plenty of practice questions complete answers an appendices With this book you will be well prepared for your exams and boost your performance CONTENTS 1 Writing the Formulae of Ionic Compounds 2 Formula Masses 3 Mass and Percent Compositions 4 The Mole and Mass 5 The Mole and Number of Particles 6 The Mole and Concentration 7 The Mole and Molar Volume 8 Empirical and Molecular Formulae 9 Chemical Equations 10 Calculations Based on Chemical Equations 11 Limiting Reagent 12 Gas Volume Volume Stoichiometry 13 Percent Yield Answers to Practice Problems Appendices You Can Do Chemistry Sunil Tanna, 2018-11-19 A comprehensive guide to performing mole and stoichiometric calculations with numerous examples as well as questions and answers Covers calculations relating to solids solutions gases and electrolysis plus as limiting and excess reactants chemical yields atom economy and much more Fully up to date with the last international standards including the revised definition of mole which was agreed on November 16th 2018 **Chemistry: 1001 Practice Problems For Dummies (+ Free Online Practice)** Heather Hattori, Richard H. Langley, 2022-06-08 Practice your way to a better grade in your Chemistry class

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