Name:



## **Balancing Chemical Equations**

Directions: Balance each of the following chemical equations.

- 1)  $C_{10}H_{22} + C_{02} \rightarrow C_{02} + C_{02}H_{20}$
- 2) \_\_\_ Al(OH)<sub>3</sub> + \_\_\_ HBr → \_\_\_ AlBr<sub>3</sub> + \_\_\_ H<sub>2</sub>O
- 3) \_\_\_CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> + \_\_\_O<sub>2</sub> → \_\_\_CO<sub>2</sub> + \_\_\_H<sub>2</sub>O
- 4) \_\_C<sub>3</sub>H<sub>8</sub> + \_\_O<sub>2</sub> → \_\_CO<sub>2</sub> + \_\_H<sub>2</sub>O
- 5) \_\_ Li + \_\_ AlO<sub>3</sub> → \_\_ LICI + \_\_ Al
- 6) \_\_\_C<sub>2</sub>H<sub>6</sub> + \_\_\_O<sub>2</sub> → \_\_\_CO<sub>2</sub> + \_\_\_H<sub>2</sub>O
- 7) \_\_\_NH4OH + \_\_\_H2O4 → \_\_\_(NH4)3PO4 + \_\_\_H2O
- 8) \_\_\_ Rb + \_\_\_ P → \_\_\_ Rb<sub>3</sub>P
- 9) \_\_CH<sub>4</sub> + \_\_O<sub>2</sub> → \_\_CO<sub>2</sub> + \_\_H<sub>2</sub>O
- 10) \_\_\_ Al(OH)<sub>3</sub> + \_\_\_ H<sub>2</sub>SO<sub>4</sub> → \_\_\_ Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> + \_\_\_ H<sub>2</sub>O
- 11) \_\_\_ Na + \_\_\_ Cl₂ → \_\_\_ NaCl
- 12) \_\_ Rb + \_\_ S<sub>8</sub> → \_\_ Rb<sub>2</sub>S
- 13) \_\_\_  $H_3PO_4$  + \_\_\_  $Ca(OH)_2$   $\rightarrow$  \_\_\_  $Ca_3(PO_4)_2$  + \_\_\_  $H_2O$
- 14) \_\_\_ NH<sub>3</sub> + \_\_\_ HCl → \_\_\_ NH<sub>4</sub>Cl
- 15) \_\_ Li + \_\_ H₂O → \_\_ LiOH + \_\_ H₂
- 16) \_\_\_ Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> + \_\_\_ SiO<sub>2</sub> + \_\_\_ C → \_\_\_ CaSiO<sub>3</sub> + \_\_\_ CO + \_\_\_ P
- 17) \_\_\_NH<sub>3</sub> + \_\_\_O<sub>2</sub> → \_\_\_N<sub>2</sub> + \_\_\_H<sub>2</sub>O
- 18) \_\_\_ FeS<sub>2</sub> + \_\_\_ O<sub>2</sub> → \_\_\_ Fe<sub>2</sub>O<sub>3</sub> + \_\_\_ SO<sub>2</sub>

# **Chemfiesta Show How To Balancing Chemical Equations**

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## **Chemfiesta Show How To Balancing Chemical Equations:**

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