

Stoichiometry Practice Worksheet

Solve the following stoichiometry grams-grams problems:

- 1) Using the following equation: $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

How many grams of sodium sulfate will be formed if you start with 200 grams of sodium hydroxide and you have an excess of sulfuric acid?

- 2) Using the following equation: $\text{Pb}(\text{SO}_4)_2 + 4 \text{LiNO}_3 \rightarrow \text{Pb}(\text{NO}_3)_4 + 2 \text{Li}_2\text{SO}_4$

How many grams of lithium nitrate will be needed to make 250 grams of lithium sulfate, assuming that you have an adequate amount of lead (IV) sulfate to do the reaction?

Limiting Reagent Worksheet

For the following reactions, find the following:

- a) Which of the reagents is the limiting reagent?
b) What is the maximum amount of each product that can be formed?
c) How much of the other reagent is left over after the reaction is complete?

- 3) Consider the following reaction:



Answer the questions above (a,b,c), assuming we started with 30 grams of ammonium nitrate and 50 grams of sodium phosphate.

- 4) Consider the following reaction:



Answer the questions above (a,b,c) assuming we start with 100 grams of calcium carbonate and 45 grams of iron (III) phosphate.

5) When copper (II) chloride reacts with sodium nitrate, copper (II) nitrate and sodium chloride are formed.

- a) Write the balanced equation for the reaction given above:
b) If 15 grams of copper (II) chloride react with 20 grams of sodium nitrate, how much sodium chloride can be formed?
c) What is the limiting reagent for the reaction in #2? _____
d) How much of the nonlimiting reagent is left over in this reaction?

Chemfiesta Stoichiometry Lab Answers

David Callaghan



Chemfiesta Stoichiometry Lab Answers:

A Stoichiometry Unit David Callaghan, 2004

Stoichiometry Unit Project Luann Marie Decker, 1998

This book delves into Chemfiesta Stoichiometry Lab Answers. Chemfiesta Stoichiometry Lab Answers is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chemfiesta Stoichiometry Lab Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Chemfiesta Stoichiometry Lab Answers
- Chapter 2: Essential Elements of Chemfiesta Stoichiometry Lab Answers
- Chapter 3: Chemfiesta Stoichiometry Lab Answers in Everyday Life
- Chapter 4: Chemfiesta Stoichiometry Lab Answers in Specific Contexts
- Chapter 5: Conclusion

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The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chemfiesta Stoichiometry Lab Answers.

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