

## WORD EQUATIONS

Name \_\_\_\_\_

Write and balance the following chemical equations.

1. Hydrogen plus oxygen yields water.



2. Nitrogen plus hydrogen yields ammonia.



3. Aluminum bromide plus chlorine yields aluminum chloride and bromine.



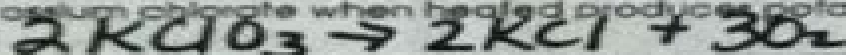
4. Hydrochloric acid plus sodium hydroxide yields sodium chloride plus water.



5. Iron plus lead (II) sulfate react forming iron (II) sulfate plus lead.



6. Potassium chlorate when heated produces potassium chloride plus oxygen gas.



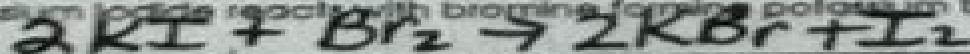
7. Sulfuric acid decomposes to form sulfur trioxide gas plus water.



8. Sodium oxide combines with water to make sodium hydroxide.



9. Potassium iodide reacts with bromine forming potassium bromide plus iodine.



10. Sodium phosphate reacts with calcium nitrate to produce sodium nitrate plus calcium phosphate.



11. Zinc reacts with iron (III) chloride yielding zinc chloride plus iron precipitate.



12. Ammonium carbonate and magnesium sulfate react to yield ammonium sulfate plus magnesium carbonate.



13. Phosphoric acid plus calcium hydroxide react forming solid calcium phosphate plus water.



14. Aluminum plus oxygen gas form aluminum oxide under certain conditions.



15. Nitrogen gas plus oxygen gas react and form dinitrogen pentoxide.



# Chemfiesta Word Equations Answer Key

**Klaus U. Schulz**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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## **Table of Contents Chemfiesta Word Equations Answer Key**

1. Understanding the eBook Chemfiesta Word Equations Answer Key
  - The Rise of Digital Reading Chemfiesta Word Equations Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemfiesta Word Equations Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemfiesta Word Equations Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemfiesta Word Equations Answer Key
  - Personalized Recommendations
  - Chemfiesta Word Equations Answer Key User Reviews and Ratings
  - Chemfiesta Word Equations Answer Key and Bestseller Lists
5. Accessing Chemfiesta Word Equations Answer Key Free and Paid eBooks
  - Chemfiesta Word Equations Answer Key Public Domain eBooks
  - Chemfiesta Word Equations Answer Key eBook Subscription Services
  - Chemfiesta Word Equations Answer Key Budget-Friendly Options

6. Navigating Chemfiesta Word Equations Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemfiesta Word Equations Answer Key Compatibility with Devices
  - Chemfiesta Word Equations Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemfiesta Word Equations Answer Key
  - Highlighting and Note-Taking Chemfiesta Word Equations Answer Key
  - Interactive Elements Chemfiesta Word Equations Answer Key
8. Staying Engaged with Chemfiesta Word Equations Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemfiesta Word Equations Answer Key
9. Balancing eBooks and Physical Books Chemfiesta Word Equations Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemfiesta Word Equations Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemfiesta Word Equations Answer Key
  - Setting Reading Goals Chemfiesta Word Equations Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemfiesta Word Equations Answer Key
  - Fact-Checking eBook Content of Chemfiesta Word Equations Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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