Name: _____ Date: ____

Homework: Writing and Balancing Chemical Equations

Mg + O₂ → MgO

When the above equation is balanced, what would be the coefficient on oxygen gas?
 A 1 B 2 C 3 D 4

$F_2 + S_8 \rightarrow S_2F_{10}$

When the above equation is balanced, what would be the coefficient on fluorine gas?
 B 10
 C 15
 D 20

$O_2 + N_2 \rightarrow N_3O_5$

3. When the above equation is balanced, what would be the coefficient on nitrogen gas?

A 2 B 3 C 4 D 5

$K + H_2O \rightarrow KOH + H_2$

4. When the above equation is balanced, what would be the coefficient on potassium hydroxide?
A 1
B 2
C 3
D 4

5. Balance the following chemical equations from scratch:

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