

Stoichiometry Practice Worksheet

Balancing Equations and Simple Stoichiometry

Balance the following equations:

- 1) $\text{___ N}_2 + \text{___ F}_2 \rightarrow \text{___ NF}_3$
- ✓ 2) $\text{___ C}_6\text{H}_{10} + \text{___ O}_2 \rightarrow \text{___ CO}_2 + \text{___ H}_2\text{O}$
- ✓ 3) $\text{___ HBr} + \text{___ KHCO}_3 \rightarrow \text{___ H}_2\text{O} + \text{___ KBr} + \text{___ CO}_2$
- ✓ 4) $\text{___ GaBr}_3 + \text{___ Na}_2\text{SO}_3 \rightarrow \text{___ Ga}_2(\text{SO}_3)_3 + \text{___ NaBr}$
- 5) $\text{___ SnO} + \text{___ NF}_3 \rightarrow \text{___ SnF}_2 + \text{___ N}_2\text{O}_3$

Solve the following stoichiometry grams-grams problems:

- ✓ 6) Using the following equation:



How many grams of sodium sulfate will be formed if you start with 200 grams of sodium hydroxide and you have an excess of sulfuric acid?

↗ Using the following equation:



How many grams of lithium nitrate will be needed to make 250 grams of lithium sulfate, assuming that you have an adequate amount of lead (IV) sulfate to do the reaction?

Chemfiesta Stoichiometry Practice Answer Key

Ian Guch



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