

WORD EQUATIONS

Name _____

Write and balance the following chemical equations.

1. Hydrogen plus oxygen yields water.



2. Nitrogen plus hydrogen yields ammonia.



3. Aluminum bromide plus chlorine yields aluminum chloride and bromine.



4. Hydrochloric acid plus sodium hydroxide yields sodium chloride plus water.



5. Iron plus lead (II) sulfate react forming iron (II) sulfate plus lead.



6. Potassium chlorate when heated produces potassium chloride plus oxygen gas.



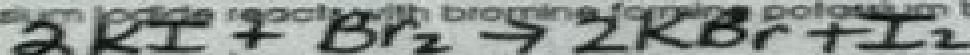
7. Sulfuric acid decomposes to form sulfur trioxide gas plus water.



8. Sodium oxide combines with water to make sodium hydroxide.



9. Potassium iodide reacts with bromine forming potassium bromide plus iodine.



10. Sodium phosphate reacts with calcium nitrate to produce sodium nitrate plus calcium phosphate.



11. Zinc reacts with iron (III) chloride yielding zinc chloride plus iron precipitate.



12. Ammonium carbonate and magnesium sulfate react to yield ammonium sulfate plus magnesium carbonate.



13. Phosphoric acid plus calcium hydroxide react forming solid calcium phosphate plus water.



14. Aluminum plus oxygen gas form aluminum oxide under certain conditions.



15. Nitrogen gas plus oxygen gas react and form dinitrogen pentoxide.



Chemfiesta Word Equations Answers

Crispin Collins



Chemfiesta Word Equations Answers:

Coordinating Procedural and Conceptual Knowledge to Make Sense of Word Equations: Understanding the Complexity of a 'simple' Chemical Task at the Learner's Resolution Keith Stephen Taber, Pat Bricheno, 2009 Abstract This paper discusses the conceptual demands of an apparently straightforward task set to secondary level students completing chemical word equations with a single omitted term Chemical equations are of considerable importance in chemistry and school students are expected to learn to be able to write and interpret them However it is recognized that many students find them challenging The present paper explores students accounts of their attempts to identify the missing terms to illuminate why working with chemical word equations is so challenging from the learner's perspective 300 secondary age students responded to a 5 item exercise based on chemicals and types of reactions commonly met at school level For each item they were asked to identify the missing term in a word equation and explain their answers This provided a database containing more than a thousand student accounts of their rationales Analysis of the data led to the identification of seven main classes of

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