

Writing and Balancing Equations Worksheet

STO.1	Balance a chemical equation.
STO.2	Identify the parts of a chemical equation.
RXN.1	Describe a chemical reaction using words and symbolic equations.

For each of the following problems, write complete chemical equations to describe the chemical process taking place. Balance the equations.

- 1) When lithium hydroxide pellets are added to a solution of sulfuric acid (H_2SO_4), lithium sulfate and water are formed.
- 2) Magnesium reacts with sodium fluoride to produce magnesium fluoride and elemental sodium.
- 3) If a copper coil is placed into a solution of silver nitrate, silver crystals form and copper (I) nitrate is generated.
- 4) When crystalline $\text{C}_6\text{H}_{12}\text{O}_6$ is burned in oxygen, carbon dioxide and water vapor are formed.
- 5) Calcium carbonate combines with hydrochloric acid (HCl) to produce calcium chloride, water and carbon dioxide gas.

Balance the equations below:

- 1) _____ N_2 + _____ $\text{H}_2 \rightarrow$ _____ NH_3
- 2) _____ $\text{KClO}_3 \rightarrow$ _____ KCl + _____ O_2
- 3) _____ NaCl + _____ $\text{F}_2 \rightarrow$ _____ NaF + _____ Cl_2
- 4) _____ H_2 + _____ $\text{O}_2 \rightarrow$ _____ H_2O
- 5) _____ $\text{Pb}(\text{OH})_2$ + _____ $\text{HCl} \rightarrow$ _____ H_2O + _____ PbCl_2
- 6) _____ AlBr_3 + _____ $\text{K}_2\text{SO}_4 \rightarrow$ _____ KBr + _____ $\text{Al}_2(\text{SO}_4)_3$
- 7) _____ CH_4 + _____ $\text{O}_2 \rightarrow$ _____ CO_2 + _____ H_2O
- 8) _____ C_3H_8 + _____ $\text{O}_2 \rightarrow$ _____ CO_2 + _____ H_2O
- 9) _____ C_6H_{18} + _____ $\text{O}_2 \rightarrow$ _____ CO_2 + _____ H_2O
- 10) _____ FeCl_3 + _____ $\text{NaOH} \rightarrow$ _____ $\text{Fe}(\text{OH})_3$ + _____ NaCl
- 11) _____ P + _____ $\text{O}_2 \rightarrow$ _____ P_2O_5

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Slabaugh



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