

Name _____

BONDING



Directions: Use the chart below and your periodic table to answer the questions that follow.

PART I: COMPARING ELEMENTS

Group One and Group Two both show different compounds made by bonding different elements together. There are two types of bonding in food science, as shown in these two groupings. Let's take a closer look at the elements in each compound to see what makes a compound IONIC or COVALENT.

IONIC BONDS	COVALENT BONDS
NaCl (Sodium Chloride)	H ₂ O (Dihydrogen Monoxide)
MgO (Magnesium Oxide)	CH ₄ (Methane)
Li ₂ O (Lithium Oxide)	CO ₂ (Carbon Dioxide)
KF (Potassium Fluoride)	HF (Hydrofluoric Acid)
FeBr ₃ (Iron III Bromide)	NH ₃ (Ammonia)
CaCl ₂ (Calcium Chloride)	NO ₂ (Nitrogen Dioxide)
NiI ₂ (Nickel Iodide)	C ₆ H ₁₂ O ₆ (Glucose)
BaS (Barium Sulfide)	CF ₄ (Carbon Tetrafluoride)

1. List out the FIRST element in each of the GROUP ONE compounds.
2. What do you notice about the FIRST element that makes up each compound in GROUP ONE?
3. What side of the periodic table do you find these FIRST elements on?
4. List out the SECOND element in each of the GROUP ONE compounds.
5. What do you notice about the SECOND element that makes up each compound in GROUP ONE?
6. What side of the periodic table do you find these SECOND elements on?

BIG IDEA #1: When a _____ and a _____ come together, they make an IONIC BOND.

7. What do you notice about ALL of the elements in GROUP TWO?
8. What side of the periodic table do you find ALL of these elements on?

Chemical Bonding Pogil Activity 1 Answers

Gary Wulfsberg

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1. Understanding the eBook Chemical Bonding Pogil Activity 1 Answers
 - The Rise of Digital Reading Chemical Bonding Pogil Activity 1 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Bonding Pogil Activity 1 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Bonding Pogil Activity 1 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Bonding Pogil Activity 1 Answers
 - Personalized Recommendations
 - Chemical Bonding Pogil Activity 1 Answers User Reviews and Ratings
 - Chemical Bonding Pogil Activity 1 Answers and Bestseller Lists
5. Accessing Chemical Bonding Pogil Activity 1 Answers Free and Paid eBooks
 - Chemical Bonding Pogil Activity 1 Answers Public Domain eBooks
 - Chemical Bonding Pogil Activity 1 Answers eBook Subscription Services
 - Chemical Bonding Pogil Activity 1 Answers Budget-Friendly Options
6. Navigating Chemical Bonding Pogil Activity 1 Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Chemical Bonding Pogil Activity 1 Answers Compatibility with Devices
 - Chemical Bonding Pogil Activity 1 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Chemical Bonding Pogil Activity 1 Answers
 - Highlighting and Note-Taking Chemical Bonding Pogil Activity 1 Answers
 - Interactive Elements Chemical Bonding Pogil Activity 1 Answers
8. Staying Engaged with Chemical Bonding Pogil Activity 1 Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Bonding Pogil Activity 1 Answers
9. Balancing eBooks and Physical Books Chemical Bonding Pogil Activity 1 Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Bonding Pogil Activity 1 Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemical Bonding Pogil Activity 1 Answers
- Setting Reading Goals Chemical Bonding Pogil Activity 1 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemical Bonding Pogil Activity 1 Answers
- Fact-Checking eBook Content of Chemical Bonding Pogil Activity 1 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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