

Name _____ Date _____ Per _____

WORKSHEET: Chemical Bonding – Ionic & Covalent!
REMEMBER...
Ionic Bond

between a Metal and Non-Metal

(M + NM)

Covalent Bond

between a Non-Metal and Non-Metal

(NM + NM)

PART 1: Determine if the elements in the following compounds are metals or non-metals. Describe the type of bonding that occurs in the compound.

| Compound | Element 1 (metal or non-metal?) | Element 2 (metal or non-metal?) | Bond Type |
|--------------------------------|------------------------------------|------------------------------------|-----------|
| NO ₂ | N = non-metal | O = non-metal | covalent |
| NaCl | | | |
| SO ₂ | | | |
| PI ₃ | | | |
| MgBr ₂ | | | |
| CaO | | | |
| H ₂ O | | | |
| K ₂ O | | | |
| AlF ₃ | | | |
| O ₂ | | | |
| CuCl ₂ | | | |
| NO ₂ | | | |
| CO ₂ | | | |
| HF | | | |
| Rb ₂ S | | | |
| NBr ₃ | | | |
| Fe ₂ O ₃ | | | |
| CCl ₄ | | | |

Chemical Bonding Practice Answer Key

Claire Aldridge, Karl Lee



Chemical Bonding Practice Answer Key:

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