

Chapter 8: Chemical Reactions

Focus of this section: *Chemical Equations*
Using balanced equations to solve problems

Chemical equations: represent chemical "sentences"

-- Read as a sentence:



"Two methanol molecules react with three oxygen molecules to produce (yield, form) 2 two carbon dioxide molecules and four water molecules"

-- **Reactant:** any species to the left of the arrow
(consumed)

-- **Product:** any species to the right of the arrow (formed)

-- **State symbols:** (s) solid (l) liquid
 (aq) water solution (g) gas

Balancing Chemical Equations:

- **Balanced:** same number and kind of atoms on each side of the equation (conservation of mass)
- (stoichiometric) coefficients: numbers before formulas



Chemical Equations And Reactions Chapter 8

Ralph Thomas Myers



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