

Balancing & Reactions Test SCH3U

Name: _____

 K T C A Assessment of Learning (Product)
10 20 5 4

Multiple Choice: Circle the most correct answer. (10K)

1. Which equation is improperly balanced?

- A) $2 \text{KNO}_3 \rightarrow 2 \text{KNO}_2 + \text{O}_2$ B) $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$
 C) $2 \text{Na}_2\text{O}_2 + 2 \text{H}_2\text{O} \rightarrow 4 \text{NaOH} + \text{O}_2$ D) $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$

2. In every balanced chemical equation, each side of the equation has the same number of ____.

- A) atoms of each element B) molecules C) Moles D) coefficients

3. Which metal would not displace lead, Pb, in a lead nitrate, $\text{Pb}(\text{NO}_3)_2$, solution?

- A) copper, Cu B) magnesium, Mg C) potassium, K D) lithium, Li E) none of the above

4. Which of the following are decomposition reactions?



- A) 1, 2, and 3 B) 4 only C) 1, 2, 3, and 4 D) 2 and 3 E) 2, 3, and 4

5. Predict the product in the synthesis reaction of Al with N_2 .

- A) AlN B) Al_3N C) AlN_2 D) Al_3N_2 E) AlN_3

6. What is true of the oxide formed when magnesium reacts with oxygen?

- A) Its formula is Mg_2O C) It is molecular
 B) It is basic D) It is amphoteric

7. Oxides of the active metals combine with acid to form _____.

- A) H_2 gas B) metal hydrides C) water and a salt D) oxygen gas E) metal hydroxides

8. Which one of the following compounds would produce an acidic solution when dissolved in water?

- A) Na_2O B) CaO C) MgO D) CO_2 E) SeO

9. In a double-displacement reaction, the ____.

- A) products are always molecular B) reactants are usually two ionic compounds
 C) reactants are two elements D) products are a new element and a new compound

Chemical Equations And Reactions Test Answers

Arshad Iqbal



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