

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- | | |
|---------------|---|
| <u>yes</u> | 1. A tomato smells rotten. |
| <u>no</u> | 2. A drinking glass breaks into smaller pieces. |
| <u>no</u> | 3. A piece of ice melts. |
| <u>yes</u> | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u> | 5. Candle wax burns. |
| <u>yes/no</u> | 6. Molten candle wax solidifies. |
| <u>yes</u> | 7. Green leaves turn yellow and red as the seasons change. |
| <u>yes</u> | 8. Baking powder produces a gas that makes a cake rise. |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

| | | | |
|----------|-----------|-----|------|
| arrow | plus sign | (s) | (l) |
| reactant | product | (g) | (aq) |

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

Chemical Reactions Study Guide Reviewing Vocabulary Answers

Tyler L. Jackson



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Macmillan/McGraw-Hill Math: Teacher ed., v. 2 ,2004

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Teaching Science for Understanding James Joseph Gallagher,2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities

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The Practice of Chemistry Donald J. Wink,Sharon Fetzer-Gislason,Sheila McNicholas,2003-03 Students can t do chemistry if they can t do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book s unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink [http www bfwpub com/pdfs/wink/POCPowerPoint_Final.ppt](http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt) 832KB

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Biology Joseph S. Levine,Kenneth R. Miller,1998

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Table of Contents Chemical Reactions Study Guide Reviewing Vocabulary Answers

1. Understanding the eBook Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - The Rise of Digital Reading Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Personalized Recommendations
 - Chemical Reactions Study Guide Reviewing Vocabulary Answers User Reviews and Ratings
 - Chemical Reactions Study Guide Reviewing Vocabulary Answers and Bestseller Lists
5. Accessing Chemical Reactions Study Guide Reviewing Vocabulary Answers Free and Paid eBooks
 - Chemical Reactions Study Guide Reviewing Vocabulary Answers Public Domain eBooks
 - Chemical Reactions Study Guide Reviewing Vocabulary Answers eBook Subscription Services
 - Chemical Reactions Study Guide Reviewing Vocabulary Answers Budget-Friendly Options
6. Navigating Chemical Reactions Study Guide Reviewing Vocabulary Answers eBook Formats

- ePub, PDF, MOBI, and More
- Chemical Reactions Study Guide Reviewing Vocabulary Answers Compatibility with Devices
- Chemical Reactions Study Guide Reviewing Vocabulary Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Highlighting and Note-Taking Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Interactive Elements Chemical Reactions Study Guide Reviewing Vocabulary Answers
- 8. Staying Engaged with Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Reactions Study Guide Reviewing Vocabulary Answers
- 9. Balancing eBooks and Physical Books Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Reactions Study Guide Reviewing Vocabulary Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Setting Reading Goals Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Fact-Checking eBook Content of Chemical Reactions Study Guide Reviewing Vocabulary Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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