

### Unit 4 Covalent Bonding Webquest

#### **Activity #1- Introduction to Covalent Bonding**

Open [Chemical Bonding](#). Scroll down to the heading "Covalent Bonding."

1. As opposed to \_\_\_\_\_ bonding in which a complete transfer of electrons occurs, \_\_\_\_\_ bonding occurs when two or more elements \_\_\_\_\_ electrons. Covalent bonding occurs because the atoms in the compound have a similar tendency for electrons (generally to \_\_\_\_\_ electrons). This most commonly occurs when two \_\_\_\_\_ bond together. Because both of the nonmetals will want to \_\_\_\_\_ electrons, the elements involved will share electrons in an effort to \_\_\_\_\_ their valence shells.
2. How many valence electrons are in one atom of hydrogen?
3. How many valence electrons does hydrogen need to have a full first shell?
4. How does the hydrogen atom "pick up" another electron?
5. What compound does hydrogen form?
6. How do hydrogen atoms make a covalent bond?
7. Visit the simulation, [Covalent bonding between hydrogen atoms](#). Describe or draw what you see.

# Chemical Bonding Webquest Answer Key

**RS Peters**

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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