BALANCING CHEMICAL EQUATIONS

Name GROUP

Rewrite and balance the equations below.

1. N, + H, - NH, Nz + 3H2 - 2NH3

2. KClO<sub>3</sub> → KCl + O<sub>2</sub> 2 KClO<sub>3</sub> → 2 KCl + 3 O<sub>2</sub>

3. NaCI + F2 -> NaF + CI2 2NaCO + F2 -> ZNaF + Cl2

4. H, +0, - HO 2H2+ 02 -> 2H20

5. AgNO<sub>3</sub> + MgCl<sub>2</sub> → AgCl + Mg(NO<sub>3</sub>)<sub>2</sub> 2AgNO<sub>3</sub> + MgCl<sub>2</sub> → ZAgOl+

6. AIBr3 + K2SO4 → KBr + AL(SO4)3 2 ALBr3+ 3 K2SO4 → 6 KBr + Al2(SO4)3

7. CH, + O2 + CO2 + HO CH4+ 202 - CO2+ 2420

8. C3H8 + O2 → CO2 + H2O C3H8+ 5O2 → 3CO2+4H2O

9. C<sub>8</sub>H<sub>18</sub> + O<sub>2</sub> → CO<sub>2</sub> + H<sub>2</sub>O C<sub>8</sub>H<sub>18</sub> + = O<sub>2</sub> → 8CO<sub>2</sub> + 9H<sub>2</sub>O

10. FOCI, + NOOH → FO(OH), + NOCI\_ FECEX + 3 NOOH → FECOH), +3NOCE

11. P + O2 - P2O6 4P + 502 -> 2P2OF

12. No + H20 → NOOH + H2 2Na+ 2 H20 → 2NaOH + H2

13. Ago - Ag + O2 2Ag20 -> 4Ag + O2

14.  $S_8 + O_2 \rightarrow SO_3$   $S_8 + 12.0 = 350_3$ 

15. CO2 + H20 → C6H12O6 + O2 6CO2 + 6H2O → C6H12O6+ 6O2

16. K + MgBr<sub>2</sub> → KBr + Mg 2 K + MgBr, → 2 KBr + Mg

17. HCI + CaCO, → CaCl, + H2O + CO, 2HCI + CaCO, → Cacl+ H1O + COL

# **Chemical Reaction And Equations Lab Key**

**Karin Nielsen-Saines** 

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Lab Manual for General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives Essentials of General Organic and Biochemistry captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession. To browse a sample chapter view sample ChemCasts and more visit www whfreeman com gob Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02 **Prentice Hall Physical Science Concepts in Action Program** Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

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web pedigrees are a useful method to model the inheritance patterns of traits within families by using symbols and analyzing phenotypes we can infer genotypes and understand the

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web pedigree problems the components of a pedigree squares are used to indicate males in a family circles are used to indicate females if the individual is affected by the trait