

EXPERIMENT 4: CHEMICAL REACTIONS

Analysis

Table 1 - Qualitative Observations

React ion	Balanced chemical equation	Observations
A	$\text{CaCl}_2(\text{aq}) + \text{Na}_3\text{PO}_4(\text{aq}) \rightarrow$	
B	$\text{CuCO}_3(\text{s}) \rightarrow$	
C	$\text{Zn}(\text{s}) + \text{HCl}(\text{aq}) \rightarrow$	
D	$\text{HCl}(\text{aq}) + \text{Na}_2\text{CO}_3(\text{aq}) \rightarrow$	
E	$\text{Sr}(\text{OH})_2(\text{s}) + \text{NH}_4\text{Cl}(\text{s}) \rightarrow \text{SrCl}_2(\text{aq}) + \text{NH}_3(\text{g}) + \text{H}_2\text{O}(\text{aq})$	
F	$\text{Zn}(\text{s}) + \text{CuSO}_4(\text{aq}) \rightarrow$	
G	$\text{HCl}(\text{aq}) + \text{NaOH}(\text{aq}) \rightarrow$	
H	$\text{Mg}(\text{s}) + \text{O}_2(\text{g}) \rightarrow$	

Chemical Reactions And Qualitative Analysis Answer Key

Annelies Wilder-Smith



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General Chemistry with Qualitative Analysis Ralph H. Petrucci,Robert K. Wismer,1983 **Oswaal JEE (Main) 22**

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