

Types of Chemical Reactions Worksheet #1

H/Chemistry

For each of the following equations, identify what kind of reaction it represents: double replacement, single replacement, decomposition, or synthesis (composition).

- 1) $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ **Synthesis**
 - 2) $\text{CaI}_2 + \text{Cl}_2 \rightarrow \text{CaCl}_2 + \text{I}_2$ **Single Replacement**
 - 3) $3\text{KOH} + \text{AlCl}_3 \rightarrow \text{Al(OH)}_3 + 3\text{KCl}$ **Double Replacement**
 - 4) $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$ **Synthesis**
 - *5) $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$ **Double Replacement**
 - 6) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ **Decomposition**
 - 7) $3\text{CuSO}_4 + 2\text{Al} \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{Cu}$ **Single Replacement**
 - 8) $\text{Na}_2\text{S} + 2\text{AgNO}_3 \rightarrow 2\text{NaNO}_3 + \text{Ag}_2\text{S}$ **Double Replacement**
 - 9) $3\text{H}_2 + \text{N}_2 \rightarrow 2\text{NH}_3$ **Synthesis**
 - *10) $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$ **Single Replacement**
- *Remember: "H₂O" can also be thought of as "HOH"

Complete each of the following equations as needed to make it the type of reaction indicated. Be sure to write each formula correctly.

- 11) Double replacement: $\text{Na}_2\text{CrO}_4 + \text{PbCl}_2 \rightarrow 2\text{NaCl} + \text{PbCrO}_4$
- 12) Single replacement: $\text{Cl}_2 + 2\text{NaBr} \rightarrow 2\text{NaCl} + \text{Br}_2$
- 13) Decomposition: $\text{Mg(ClO}_3)_2 \rightarrow \text{MgCl}_2 + 3\text{O}_2$
- 14) Synthesis: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- 15) Double replacement: $3\text{Ca(OH)}_2 + 2\text{FeCl}_3 \rightarrow 2\text{Fe(OH)}_3 + 3\text{CaCl}_2$
- 16) Single replacement: $\text{Fe} + \text{Cu(NO}_3)_2 \rightarrow \text{Fe(NO}_3)_2 + \text{Cu}$ [Assume Fe²⁺]
- 17) Decomposition: $2\text{Hg}_2\text{O} \rightarrow 4\text{Hg} + \text{O}_2$
- 18) Synthesis: $\text{S} + \text{O}_2 \rightarrow \text{SO}_2$
- 19) Double replacement: $\text{AgNO}_3 + \text{KI} \rightarrow \text{AgI} + \text{KNO}_3$
- 20) Single replacement: $\text{Cu} + 2\text{AgNO}_3 \rightarrow 2\text{Ag} + \text{Cu(NO}_3)_2$ [Copper (II) is used here]

Chemical Reactions Answers

Amar Bhutani



Chemical Reactions Answers:

Balancing Chemical Equations Workbook Kraaya Publishing, 2021-02-21 Chemical Reactions to Balance Workbook This chemistry balancing equations practice workbook contains 250 non balanced chemical equations Begin with 2 terms problems Work your way up to 6 terms problems This is the perfect workbook to increase chemistry balancing skills for beginners Table of contents How To Balance A Chemical Equation Chemical Equations To Balance Correct Answers Book features Non repetitive equations Include all reactions types synthesis combustion decomposition Use it now and develop instant recall of balancing equations Enjoy the challenge *Balancing Chemical Equations Worksheet* Crispin Collins, 2020-09-12 Struggling with balancing chemical reaction Balancing chemical equations can look intimidating for lot of us The good news is that practice makes perfect Master balancing skill with this workbook packed with hundreds of practice problems This book is for anyone who wants to master the art of balancing chemical reactions First few chapters of this book are step by step explanation of the concepts and other chapters are for practicing problems This book help students develop fluency in balancing chemical equation which provides plenty of practice Methods to solve with the explanation Total of 550 problems to solve with answer key 450 chemical reactions to practice with answer key 100 practice problems that are needed before balancing a chemical reaction with answer key Click the Buy now button to take advantage of this book to help yourself in mastering balancing skill **Chemistry Equations & Answers** Speedy Publishing, 2014-08-01 One of the most important parts of learning chemistry is simply knowing all of the equations and formulas that are used in it It is important to know the exact formatting of these equations and most classes will require a student to know them for exams It is a good idea to learn these equations and formulas with the use of a study pamphlet The pamphlet can condense all of the information so a student can memorize the equations and formulas while studying Examination Questions and Answers in Basic Anatomy and Physiology Martin Caon, 2020-08-03 This third edition provides 2900 multiple choice questions on human anatomy and physiology and some biophysical science separated into 20 chapters and 68 categories In addition there are 64 essay topics The answer to each question is accompanied by an explanation Each chapter has an introduction to set the scene for the questions to come However not all possible information is provided within these Introductions so an Anatomy and Physiology textbook is an indispensable aid to understanding the answers The textbook offers a more holistic approach to the subjects of anatomy and physiology by also including biomechanics biophysics and biochemistry The questions have been used in end of semester examinations for undergraduate anatomy and physiology courses and as such reflect the focus of these particular courses and are pitched at this level to challenge students that are beginning their training in anatomy and physiology The question and answer combinations are intended for use by teachers to select questions for their next examinations and by students when studying for an upcoming test Students enrolled in the courses for which these questions were written include nursing midwifery paramedic physiotherapy occupational therapy nutrition and dietetics health

sciences exercise science and students taking an anatomy and physiology course as an elective

Concise Chemistry class 9 icse solutions Amar Bhutani, This book includes the solutions to the questions given in the textbook ICSE Concise Chemistry Class 9 published by Selina Publications and is for March 2022 Examinations

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Balancing Chemical Equations Worksheets (Over 200 Reactions to Balance) Chris McMullen, 2016-01-12 Master the art of balancing chemical reactions through examples and practice 10 examples are fully solved step by step with explanations to serve as a

guide Over 200 chemical equations provide ample practice Exercises start out easy and grow progressively more challenging and involved Answers to every problem are tabulated at the back of the book A chapter of pre balancing exercises helps develop essential counting skills Opening chapter reviews pertinent concepts and ideas Not just for students Anyone who enjoys math and science puzzles can enjoy the challenge of balancing these chemical reactions *The Pharmacist's Handbook with State Board Examination Questions and Answers* International Correspondence Schools, 1923 [Access to Chemistry](#) Avril Higton, Mike Clemmet, Elaine Golding, Alan V Jones, 2007-10-31 This text has been specifically designed to prepare people with previously limited chemical knowledge for entrance into science related courses such as Foundation and Access courses which involve chemistry in higher education Until now there have been no texts available for use on these courses and this book fills that gap Access to Chemistry effectively forms a self study course which is split into separate modules and units covering the full spread of concepts required for those needing a basic knowledge of chemistry The material is presented in a friendly and easy to use manner which allows the student to pace their acquisition of knowledge and gain increasing confidence in order to succeed in understanding essential relevant concepts Other useful features of this book include starter diagnostic tests worked examples and self study tests with answers at the end of each unit In addition to Access or Foundation course students and their tutors to whom this book will prove essential it will have an appeal also as a revision text for those needing a refresher after a break in the subject In addition it will be of interest to members of the general public who wish to better educate themselves on chemical matters as it provides a clear and useful insight into areas such as health home chemicals business market trends and gardening **Micellar Catalysis** Mohammad Niyaz Khan, 2006-08-24 Micelles are prevalent in naturally occurring and biological catalytic reactions However it is only in recent decades that scientists have developed kinetic models clarifying how micelle mediated catalysis works at a molecular level Written by a leading expert in the field Micellar Catalysis is an in depth examination of how micelles affect reactions **Verbal Reactions - Word Scrambles With a Chemical Flavor (Medium)** Chris McMullen, Carolyn Kivett, 2011-12-09 VERBAL ReAcTiONS are a new form of word scrambles with a chemical flavor You don't need to know any science to be able to solve VERBAL ReAcTiONS puzzles but the puzzles bear a resemblance to chemical reactions EXAMPLE Es 2 S P Si 2 O N _ _ _ _ _ _ _ _ This VERBAL ReAcTiON is a word scramble consisting of one Es two S s one P one Si two O s and one N That is the word scramble contains the elements Es S S P Si O O and N Unscramble these elements to form an 8 symbol word that's why there are 8 blanks in the puzzle ANSWER For this puzzle the answer is P O S S Es Si O N possession These VERBAL ReAcTiONS resemble chemical reactions in two ways First the scrambled elements appear added together on the left of the reaction with coefficients like the number 2 in the puzzle above telling you how many of each element the solution contains and you fill in the result of the VERBAL ReAcTiON by rearranging the elements and writing them on the blanks on the right side of the reaction Secondly all of the solutions are chemical words A chemical word is a word that can be made using

symbols from the periodic table For example the chemical word POSSEsSiON is made using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es silicon Si and nitrogen N You don t need to be familiar with the periodic table to solve these problems nor do you need to know any chemistry You just need to be able to count and unscramble elements to make words This Medium volume consists of words with 6 symbols which involves familiarity with common 6 to 12 letter words A unique feature of this book is that there is a Hints section at the back separate from the Answers section for puzzlers who may be stuck and want to check just the first letter of the solution MORE EXAMPLES 1 S Ni Ge U _ _ _ _ 2 2 C N 2 I P _ _ _ _ _ 3 Ti C Cr P Y _ _ _ _ _ 4 2 C U 2 S Es _ _ _ _ _ You can find the answers below Note that this medium volume consists of chemical words with 6 symbols There are also easy and hard puzzles available in separate volumes which consist of shorter and longer chemical words ANSWERS 1 GeNiUS 2 PICNIC 3 CrYPTiC 4 SUCCEsS Student Solutions Manual Darrell Ebbing,2005-03-17 Provides solutions to odd numbered Practice Problems General Problems and Cumulative Skills Problems plus answers to Review Questions State board examination questions & answers of the United States and Canada Medical record,1912 *State Board Examination Questions & Answers of the United States and Canada* ,1912 Science Examinations ... Reports, Etc ,1904 **Conquest** ,1922 **The National Druggist** ,1922 *Scientifica* David Sang,Peter Ellis,Derek McMonagle,2004 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities **Holt Chemistry** Ralph Thomas Myers,2004

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