

Chemical Reactions

Signs of Chemical Reactions

What is a chemical reaction? _____

Chemical Formulas

A chemical formula uses _____ to identify the _____ in a compound and the _____ of each kind present in a molecule.



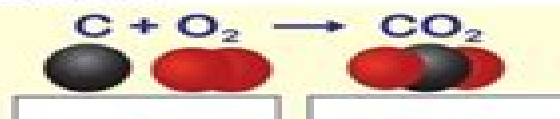
A water molecule is made up of _____ atoms of _____ bonded to _____ atom of _____.



A glucose molecule is made up of _____ atoms of _____, _____ atoms of _____, and _____ atoms of _____.

Chemical Equations

A chemical equation uses _____ and _____ to describe a chemical reaction. The **starting materials** in a chemical reaction are _____. The **resulting substances** from a reaction are _____. The _____ are always placed on the left of the arrow, and the _____ on the right of the arrow.



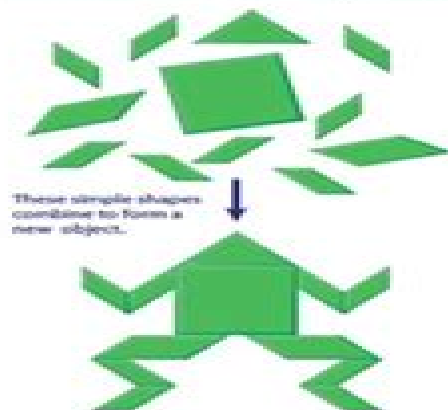
Key Vocabulary Terms

- atom
- chemical equation
- chemical formula
- chemical reaction
- compound
- decomposition reaction
- double replacement reaction
- element
- product
- reactant
- single replacement reaction
- synthesis reaction

Types of Chemical Reactions

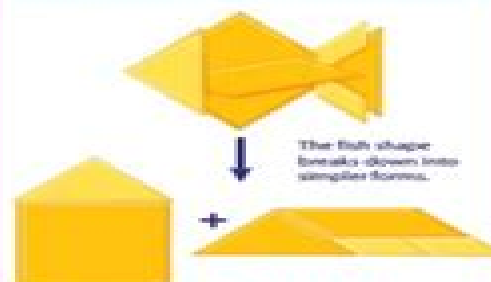
Synthesis Reaction

Describe a synthesis reaction:



Decomposition Reaction

Describe a decomposition reaction:



Replacement Reaction

Single Replacement

Describe a single replacement reaction:



Double Replacement

Describe a double replacement reaction:



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Illustrated Guide to Home Chemistry Experiments Robert Bruce Thompson, 2008-04-29 For students DIY hobbyists and science buffs who can no longer get real chemistry sets this one of a kind guide explains how to set up and use a home chemistry lab with step by step instructions for conducting experiments in basic chemistry not just to make pretty colors and stinky smells but to learn how to do real lab work Purify alcohol by distillation Produce hydrogen and oxygen gas by electrolysis Smelt metallic copper from copper ore you make yourself Analyze the makeup of seawater bone and other common substances Synthesize oil of wintergreen from aspirin and rayon fiber from paper Perform forensics tests for fingerprints blood drugs and poisons and much more From the 1930s through the 1970s chemistry sets were among the most popular Christmas gifts selling in the millions But two decades ago real chemistry sets began to disappear as manufacturers

and retailers became concerned about liability. The Illustrated Guide to Home Chemistry Experiments steps up to the plate with lessons on how to equip your home chemistry lab, master laboratory skills and work safely in your lab. The bulk of this book consists of 17 hands-on chapters that include multiple laboratory sessions on the following topics: Separating Mixtures, Solubility and Solutions, Colligative Properties of Solutions, Introduction to Chemical Reactions, Stoichiometry, Reduction/Oxidation, Redox Reactions, Acid-Base Chemistry, Chemical Kinetics, Chemical Equilibrium and Le Chatelier's Principle, Gas Chemistry, Thermochemistry and Calorimetry, Electrochemistry, Photochemistry, Colloids and Suspensions, Qualitative Analysis, Quantitative Analysis, Synthesis of Useful Compounds, Forensic Chemistry. With plenty of full-color illustrations and photos, Illustrated Guide to Home Chemistry Experiments offers introductory-level sessions suitable for a middle school or first-year high school chemistry laboratory course and more advanced sessions suitable for students who intend to take the College Board Advanced Placement AP Chemistry exam. A student who completes all of the laboratories in this book will have done the equivalent of two full years of high school chemistry lab work or a first-year college general chemistry laboratory course. This hands-on introduction to real chemistry using real equipment, real chemicals and real quantitative experiments is ideal for the many thousands of young people and adults who want to experience the magic of chemistry.

Exam Survival Guide: Physical Chemistry Jochen Vogt, 2017-02-24. A text and exercise book for physical chemistry students. This book deals with the fundamental aspects of physical chemistry taught at the undergraduate level in chemistry and the engineering sciences in a compact and practice-oriented form. Numerous problems and detailed solutions offer the possibility of an in-depth reflection of topics like chemical thermodynamics and kinetics, atomic structure and spectroscopy. Every chapter starts with a recapitulation of important background information before leading over to representative exercises and problems. Detailed descriptions systematically present and explain the solutions to the problems so that readers can carefully check their own solutions and get clear-cut introductions on how to approach similar problems systematically. The book addresses students at the upper undergraduate level as well as tutors and teachers. It is a rich source of exercises for exam preparation and can be used alongside classical textbooks. Furthermore, it can serve teachers and tutors for the conception of their lessons. Its well-thought-through presentation structure and design make the book appeal to everybody who wants to succeed with the physical chemistry lessons and exercises.

Chemistry for the IB Diploma Study and Revision Guide Christopher Talbot, Richard Harwood, 2017-07-24. Exam Board: IB Level: IB Subject: Chemistry. First Teaching: September 2014. First Exam: Summer 2016. Stretch your students to achieve their best grade with these year-round course companions providing clear and concise explanations of all syllabus requirements and topics and practice questions to support and strengthen learning. Consolidate revision and support learning with a range of exam practice questions and concise and accessible revision notes. Practise exam technique with tips and trusted guidance from examiners on how to tackle questions. Focus revision with key terms and definitions listed for each topic/sub-topic.

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