

ACS Chemistry Final (general chem 1 review) 2024/2025 Exam Questions with 100% Correct Answers | Latest Update

Density - **ANSWER** ✓✓mass / volume

Conversion from celsius to kelvin - **ANSWER** ✓✓ $K = \text{DegC} + 273.15$

Conversion from DegF to DegC - **ANSWER** ✓✓ $\text{DegC} = (5/9) (\text{DegF} - 32)$

Conversion from DegC to DegF - **ANSWER** ✓✓ $\text{DegF} = (9/5)(\text{DegC}) + 32$

Intensive properties - **ANSWER** ✓✓Independent of amount of matter examined

Extensive propertires - **ANSWER** ✓✓Dependent of amount of matter examined

Alcohol - **ANSWER** ✓✓compound in which an H atom of a hydrocarbon is replaced by an OH funxtional group (alcohols end in -ol)

Formula weight - **ANSWER** ✓✓sum of the atomic weights of each atom in its chemical formula

Percent composition - **ANSWER** ✓✓ $\% \text{ element} = (\text{number of atoms of that element})(\text{atomic weight of element}) / \text{formula weight of compound}$

Avogadro's number - **ANSWER** ✓✓ 6.02×10^{23}

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**Western Australia. Department of
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Oxford University Gazette University of Oxford,1905 *Technical Information Pilot* ,1951 **A Dictionary of Applied Chemistry** Thomas Edward Thorpe,1912 Essentials of Physical Chemistry Don Shillady,2011-07-27 At a time when U S high school students are producing low scores in mathematics and science on international examinations a thorough grounding in physical chemistry should not be considered optional for science undergraduates Based on the author s thirty years of teaching *Essentials of Physical Chemistry* merges coverage of calculus with chemist **Report on the Progress of Education in the Punjab for the Year** ,1914 **The Lancet** ,1894 *The Journal of Industrial and Engineering Chemistry* ,1913 *Survey of the School System* British Columbia. Department of Education,John Harold Putman,George Moir Weir,1925 **Pharmaceutical Journal;** ,1902 **Psychological Monographs** ,1918 Includes music

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