

S332 Chemistry
09-10 First Semester Review

Name: _____

This packet is optional, however, it will really help you review the concepts that will be on the final exam.

12 inches = 1 foot
3 feet = 1 yard
16.5 feet = 1 rod
5280 feet = 1 mile
40 rods = 1 furlong
6 feet = 1 fathom
1 gallon = 231 inch³ δ
16 ounces = 1 pint
2 pints = 1 quart
4 quarts = 1 gallon

1 kilometer = 1000 meters
1 hectometer = 100 meters
1 decameter = 10 meters
10 decimeters = 1 meter
100 centimeters = 1 meter
1000 millimeters = 1 meter
1 milliliter = 1 centimeter³

1000 milliliters = 1 liter

2.54 centimeters = 1 inch δ

English to metric conversion:

*For the most part, the equivalencies in this table are exact and therefore will not affect the number of significant figures in the answer to the problems in this chapter. The exceptions are marked with an δ .

1. Convert each of the following quantities into the required unit.
 - a. 12.75 mm to ? kilometers
 - b. 277 cm to ? meters
 - c. 81.9 cm² to ? square meters
 - d. 10.74 cm³ to ? liters
 - e. 7.48 m² to ? square inches
 - f. 3.72 L/hour to ? cubic centimeters per minute
 - g. 6.12 km/day to ? meter per second
2. A textbook measures 250. mm long, 224 mm wide, and 50.0 mm thick. It has a mass of 2.94 kg.
 - a. What is the volume of the book?
 - b. What is the density of the book in grams per cubic centimeter?
3. A gas has a density of 5.56 g/L.
 - a. What is the volume in milliliters if you have 4.17 grams of the gas in a container?

Chemistry 1 First Semester Review Packet

Yi-Tong Ma



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