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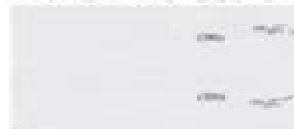
Where is the hydrophilic (attracted to water) region of the molecule?

- (A) Region 1
- (B) Region 2
- (C) Region 3
- (D) The three regions are equally hydrophilic



C

Which molecule is most soluble in water?



D

A solution of NaCl in water has a concentration of 20.5% by mass. What is the molal concentration of the solution?

Molar Mass NaCl = 58.44 g/mol

- (A) 0.205 m
- (B) 0.258 m
- (C) 3.51 m
- (D) 4.41 m

B

What is the mole fraction of water in 200 g of 89% (by mass) ethanol, C₂H₅OH?

Molar Mass C₂H₅OH = 46 g/mol

- (A) 0.11
- (B) 0.24
- (C) 0.32
- (D) 0.76

D

A mixture of 100 g of K₂Cr₂O₇ and 200 g of water is stirred at 60 degrees C until no more of the salt dissolves. The resulting solution is poured off into a separate beaker,

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Padhraic Smyth



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Table of Contents Chemistry 2 Final Exam Study Guide

1. Understanding the eBook Chemistry 2 Final Exam Study Guide
 - The Rise of Digital Reading Chemistry 2 Final Exam Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 2 Final Exam Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 2 Final Exam Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 2 Final Exam Study Guide
 - Personalized Recommendations
 - Chemistry 2 Final Exam Study Guide User Reviews and Ratings
 - Chemistry 2 Final Exam Study Guide and Bestseller Lists
5. Accessing Chemistry 2 Final Exam Study Guide Free and Paid eBooks

- Chemistry 2 Final Exam Study Guide Public Domain eBooks
 - Chemistry 2 Final Exam Study Guide eBook Subscription Services
 - Chemistry 2 Final Exam Study Guide Budget-Friendly Options
6. Navigating Chemistry 2 Final Exam Study Guide eBook Formats
- ePub, PDF, MOBI, and More
 - Chemistry 2 Final Exam Study Guide Compatibility with Devices
 - Chemistry 2 Final Exam Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Chemistry 2 Final Exam Study Guide
 - Highlighting and Note-Taking Chemistry 2 Final Exam Study Guide
 - Interactive Elements Chemistry 2 Final Exam Study Guide
8. Staying Engaged with Chemistry 2 Final Exam Study Guide
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 2 Final Exam Study Guide
9. Balancing eBooks and Physical Books Chemistry 2 Final Exam Study Guide
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 2 Final Exam Study Guide
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry 2 Final Exam Study Guide
- Setting Reading Goals Chemistry 2 Final Exam Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry 2 Final Exam Study Guide
- Fact-Checking eBook Content of Chemistry 2 Final Exam Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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