Empirical Formula of Copper Sulfide

Lab Report

Table 1 and	2 com	bined
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ble 1 and 2 combined	Trial 1	Trial 2
Mass of empty crucible (g)	9.70	111
Mass of crucible and copper (g)	9.91	The state of the s
First heating: Mass of crucible and compound (g)	11.01	
Second heating: Mass of crucible and compound (g)	4.48	No. of the last of
Third heating (only if needed): Mass of crucible and compound (g)	and though	1 13
Mass of product (g)		
Mass of copper (g)	m that one	
Mass of sulfur (g)		

Calculations

1. For each trial, determine the number of moles of copper in your compound. Show the calculation below and enter the result in Table 2.

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Hussin A.Rothana

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