- A neutral atom has the same number of electrons as protons which is also the Atomic or element number on the periodic table.
- 4. When writing electron configurations we always write them to represent (highlight one) the number of electrons in the outermost shell, or the total electrons for the neutral atom, or the number of electrons actually present. (this includes long)

#### Part 3: Electron Configurations

<ol> <li>Write the full electron configurations for each of the following a. Rhodium (Rh)</li> </ol>
b. Chloride ion (Cl <sup>2</sup> )
c. Cesium ion (Cs <sup>3+</sup> )
Write the shorthand or abbreviated electron or "Noble Gas" configuration for each of the following a. Scandium (Sc)
b. Sulfide ion (S*)

c. Aluminum ion (Al<sup>3\*</sup>)

# **Chemistry 303 Electrons In The Atom Answer Key**

Judd E. Hollander

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