

AQA Chemistry

GCSE Student calculation sheet

C4.4

Name Class Date

Questions

- 1 Calculate the percentage yield for each of the following reactions given the actual and theoretical yields.

a Zoe makes 20.5 g of copper chloride.

The maximum possible amount she could have made is 35.6 g.

.....
.....

(1 mark)

b George isolates 5.6 g of a clean, dry precipitate of iron(II) hydroxide.

The maximum possible amount he could have isolated is 15.3 g.

.....
.....

(1 mark)

c The teacher tells Aivija that she could theoretically make 1.26 mol of aspirin.

When she carries out the reaction she finds she has made 0.87 mol.

.....
.....

(1 mark)

d In the industrial process to make titanium, it is theoretically possible to extract 205 kg of titanium from 350 kg of its ore. When the process is carried out, 200 kg of titanium is extracted.

.....
.....

(1 mark)

- 2 A solution of sodium chloride reacts with silver nitrate solution to produce a white precipitate of silver chloride.

The equation for the reaction is:



- H** a Show that the maximum possible mass of AgCl that can be obtained from 58.5 g of NaCl is 143.5 g.

.....
.....
.....

(3 marks)

Chemistry 4questions Answers

JS Bruner



Chemistry 4questions Answers:

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