

CHEM 122 Exam #1 Review Questions

TRUE/FALSE. Write 'T' if the statement is true and 'F' if the statement is false.

- 1) Rates of reaction can be positive or negative. 1) _____
- 2) The relationship between the concentrations of reactants and products of a system at equilibrium is given by the law of energy. 2) _____
- 3) Le Châtelier's principle states that if a system at equilibrium is disturbed, then more reactants are produced. 3) _____

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

Chapter 15

- 4) Write a balanced reaction for which the following rate relationships are true. 4) _____

$$\text{Rate} = -\frac{1}{2} \frac{\Delta[\text{N}_2\text{O}_5]}{\Delta t} = \frac{1}{4} \frac{\Delta[\text{NO}_2]}{\Delta t} = \frac{\Delta[\text{O}_2]}{\Delta t}$$

- A) $4 \text{ NO}_2 + \text{O}_2 \rightarrow 2 \text{ N}_2\text{O}_5$
 - B) $\frac{1}{4} \text{ NO}_2 + \text{O}_2 \rightarrow \frac{1}{2} \text{ N}_2\text{O}_5$
 - C) $2 \text{ N}_2\text{O}_5 \rightarrow \text{NO}_2 + 4 \text{ O}_2$
 - D) $2 \text{ N}_2\text{O}_5 \rightarrow 4 \text{ NO}_2 + \text{O}_2$
 - E) $\frac{1}{2} \text{ N}_2\text{O}_5 \rightarrow \frac{1}{4} \text{ NO}_2 + \text{O}_2$
- 5) Given the following balanced equation, determine the rate of reaction with respect to $[\text{Cl}_2]$. If the rate of Cl_2 loss is $4.24 \times 10^{-2} \text{ M/s}$, what is the rate of formation of NO ? 5) _____



- A) $2.12 \times 10^{-2} \text{ M/s}$
 - B) $4.24 \times 10^{-2} \text{ M/s}$
 - C) $1.06 \times 10^{-1} \text{ M/s}$
 - D) $1.61 \times 10^{-2} \text{ M/s}$
 - E) $8.48 \times 10^{-2} \text{ M/s}$
- 6) The combustion of ethylene proceeds by the reaction 6) _____



When the rate of disappearance of O_2 is 0.28 M s^{-1} , the rate of appearance of CO_2 is _____ M s^{-1} .

- A) 0.84
- B) 0.19
- C) 0.093
- D) 0.56
- E) 0.42

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Didier Musso



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