

Chem 111 Final Exam

Calorimetry - Answer-science of measuring heat

Calculate amount of energy released using $E=mc(dT)$ - Answer-Energy released= (mass of water heated) \times (specific heat of water)(temp. change) specific heat=1.0 cal/gC 1 Cal=1000 cal

What is meant by small calorie "c" and food calories "C"? - Answer-"c" is a scientific calorie which is the energy required to rise the temp of 1 gram of water by 1C

"C" is 1000 scientific calories-Food Calories

When walking, an average person burns 0.453 Food Calories/ mile-pounds. How many food calories does a 150-pound person burn after walking a mile? If a peanut realizes 5.7 Food Calories/g and an average peanut weighs about 1 gram, then how many peanuts will you have to eat to walk a mile? - Answer- $\{(0.453 \text{ Food Calories})/(\text{mile-pounds})\} \times 150 \text{ lbs} \times 1 \text{ mile} = 67.95 \text{ Cal}$

$67.95 \text{ Cal} \times (1 \text{ peanut}/5.7 \text{ Food Calories}) = 12$

Define spectroscopy - Answer-study of the interactions of electromagnetic radiation with matter

Explain the difference between a line and continuous spectrum, and give an example of a source of each type - Answer-Continuous spectrum contains all frequencies within a region of the electromagnetic spectrum. e.g.- incandescent light bulb

Line spectrum- contains only certain frequencies. e.g.- Fraunhofer lines from sun

What kind of energy transitions were observed in experiment 2: spectroscopy? - Answer-valence electrons

Describe the term "white-light", and know the wavelength range (in nm). List the colors of the spectrum from highest to the lowest frequency - Answer-small band of radiation that can be seen by the human eye.

400-800 nm

ROYGBV

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