

AP Chemistry Chemical Equations Worksheet

Introduction

Help your AP Chemistry students review chemical equations with this worksheet.

Concepts

- AP chemistry review
- Chemical reactions

Discussion

The College Board has provided only a few examples of the types of questions that might be asked. (Visit the College Board Web site at www.apcentral.collegeboard.com/apc/public/courses/150180.html?type=print for examples.) AP Chemistry teachers have asked us if we have any products or worksheets to address this new format. The following list of practice chemical equations is provided as a helpful tool to assist teachers. We have compiled these questions based on our interpretation of the questions on the College Board Web site. Flinn Scientific has no additional information or guidance other than that provided by the College Board on its Web site.

Changes to the AP Chemistry Exam format for 2007 include modification to Question 4 in Section II. Previously, students were asked to write chemical equations for any five of eight given sets of chemical reactions. The new format requires students to write balanced chemical equations showing only the reacting substances (excluding spectator ions) for three different sets of reactants and to answer a short question (requiring no calculator) about each reaction. The questions are intended to test the students' understanding of the meaning of the reactions.

In each of the 38 reactions listed below, a representative question about the reaction and the answer to the question are included, along with the balanced chemical equation. We hope that you will find this worksheet helpful in preparing students for the new AP Chemistry chemical equation section.

Connecting to the National Standards

This laboratory activity relates to the following National Science Education Standards (1996):

Unifying Concepts and Processes: Grades K-12

Systems, order, and organization

Content Standards: Grades 9-12

Content Standard A: Science as Inquiry

Content Standard B: Physical Science, structure and properties of matter, chemical reactions, interactions of energy and matter

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Erin C. Amerman

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general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This

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Green Chemistry Education Paul T. Anastas,Irvin Jay Levy,Kathryn E. Parent,2009 Green Chemistry has brought about dramatic changes in the teaching of chemistry that have resulted in increased student excitement for the subject of chemistry new lecture materials new laboratory experiments and a world wide community of Green Chemistry teachers This book features the cutting edge of this advance in the teaching of chemistry **Bulletin** University of Minnesota,1924

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