

SACRED HEART CENTRAL SCHOOL (SENIOR SECONDARY)

VILLUPURAM

CHEMISTRY - 043

STD: XII

TIME: 3.00 HRS

HALF PORTION TEST (03/01/23)

CHAPTERS: 3,8,10,12&14

MAXIMUM MARKS: 70

SECTION A

The following questions are multiple-choice questions with **one** correct answer. Each question carries 1 mark. There is no internal choice in this section

- An electrochemical cell behave like an electrolytic cell when
(a) $E_{\text{cell}} = E_{\text{external}}$ (b) $E_{\text{cell}} = 0$ (c) $E_{\text{external}} > E_{\text{cell}}$ (d) $E_{\text{external}} < E_{\text{cell}}$
- Which of the following statement about transition element is not correct?
(a) They show variable oxidation states. (b) They exhibit diamagnetic and paramagnetic properties.
(c) All ions are coloured. (d) They exhibit catalytic property.
- Which of the following is most preferred to prepare alkylchloride from alcohol?
(a) Concentrated HCl (b) SOCl_2 (c) PCl_5 (d) PCl_3
- Carbonyl group undergoes
(a) Electrophilic addition reaction (b) Nucleophilic addition reactions (c) **Both** (d) None
- Which of the following reaction confirms the presence of carbonyl group ($> \text{C}=\text{O}$) in glucose?
(a) Reaction with HI (b) Reaction with hydroxylamine (c) Reaction with HCN (d) **Both (b) and (c)**
- How many coulombs are required for the oxidation of 1 mol of H_2O to O_2 ?
(a) 391000 C (b) **193000 C** (c) 319000 C (d) 390100 C
- Which one of following does not correctly represent the correct order of the property indicated against it ?
(a) $\text{Ti} < \text{V} < \text{Cr} < \text{Mn}$, increasing number of oxidation states
(b) $\text{Ti} < \text{V} < \text{Mn} < \text{Cr}$, increasing second ionisation enthalpy
(c) **$\text{Ti} < \text{V} < \text{Cr} < \text{Mn}$, increasing melting point**
(d) $\text{Ti}^{3+} < \text{V}^{3+} < \text{Cr}^{3+} < \text{Mn}^{3+}$, increasing magnetic moment
- The reagents for the following conversion is/are $\text{Br}(\text{CH}_2)_2\text{Br} \rightarrow \text{C}_2\text{H}_2$
(a) $\text{Zn}/\text{CH}_3\text{OH}$ (b) alc. KOH followed by NaNH_2 (c) aq. KOH followed by NaNH_2 (d) alc. KOH
- Fructose reduces Tollens' reagent due to
(a) Primary alcoholic group
(b) Asymmetric carbons
(c) Secondary alcoholic groups
(d) **Enolisation of fructose followed by conversion to aldehyde by base**
- Which of the following naturally occurring α - aminoacids is optically inactive?
(a) **Glycine** (b) Alanine (c) Leucine (d) Valine
- Alkyl halides undergoing nucleophilic bimolecular substitution involve
(a) retention of configuration (b) racemic mixture (c) **invert configuration** (d) formation of carbocation
- The electrode potential data are given below
 $\text{Fe}^{3+} + e^- \rightarrow \text{Fe}^{2+}$ $E^\circ = +0.77\text{V}$; $\text{Al}^{3+} + 3e^- \rightarrow \text{Al}$ $E^\circ = -1.66\text{V}$; $\text{Br}_2 + 2e^- \rightarrow 2\text{Br}^-$ $E^\circ = +1.80\text{V}$
Based on the data, the reducing power of Fe^{2+} , Al, Br^- will increase in the order.
(a) $\text{Br}^- < \text{Fe}^{2+} < \text{Al}$ (b) $\text{Fe}^{2+} < \text{Al} < \text{Br}^-$ (c) $\text{Al} < \text{Br}^- < \text{Fe}^{2+}$ (d) **$\text{Al} < \text{Fe}^{2+} < \text{Br}^-$**
- Name the type of cell which was used in Apollo space programme.
(a) Primary cell (b) dry cell (c) **fuel cell** (d) mercury cell
- Actinoids exhibit greater number of oxidation states than lanthanoids. The main reason being
(a) more energy difference between $5f$ and $6d$ than between $4f$ and $5f$ orbitals
(b) $4f$ orbitals are more diffused than the $5f$ orbitals
(c) **lesser energy difference between $5f$ and $6d$ than between $4f$ and $5d$ orbitals**
(d) more reactive nature of the actinoids than the lanthanoids.

For questions number 15 to 18, two statements are given - one labelled Assertion (A) and the other labelled Reason (R). Select the correct answer to these questions from the codes (i), (ii), (iii) and (iv) as given below:

- Both Assertion and Reason are correct statements; Reason (R) is correct explanation of Assertion (A).
- Both Assertion and Reason are correct statements. Reason (R) is not correct explanation of Assertion (A).

Chem Inquiry Practice With Transition Metals Answer Key

Mary Burnham, Carol Hurd



Chem Inquiry Practice With Transition Metals Answer Key:

Digital Learning and Teaching in Chemistry Yehudit Judy Dori, Courtney Ngai, Gabriela Szeinberg, 2023-07-12

Education is always evolving and most recently has shifted to increased online or remote learning Digital Learning and Teaching in Chemistry compiles the established and emerging trends in this field specifically within the context of learning and teaching in chemistry This book shares insights about five major themes best practices for teaching and learning digitally digital learning platforms virtual visualisation and laboratory to promote learning in science digital assessment and building communities of learners and educators The authors are chemistry instructors and researchers from nine countries contributing an international perspective on digital learning and teaching in chemistry While the chapters in this book span a wide variety of topics as a whole they focus on using technology and digital platforms as a method for supporting inclusive and meaningful learning The best practices and recommendations shared by the authors are highly relevant for modern chemistry education as teaching and learning through digital methods is likely to persist Furthermore teaching chemistry digitally has the potential to bring greater equity to the field of chemistry education in terms of who has access to quality learning and this book will contribute to that goal This book will be essential reading for those working in chemical education and teaching Yehudit Judy Dori is internationally recognised formerly Dean of the Faculty of Education of Science and Technology at the Technion Israel Institute of Technology and won the 2020 NARST Distinguished Contributions to Science Education through Research Award DCRA for her exceptional research contributions Courtney Ngai and Gabriela Szeinberg are passionate researchers and practitioners in the education field Courtney Ngai is the Associate Director of the Office of Undergraduate Research and Artistry at Colorado State University Gabriela Szeinberg serves as Assistant Dean and Academic Coordinator for the College of Arts and Sciences at Washington University in St Louis

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Scientific and Technical Aerospace Reports ,1992 Organometallic Chemistry in the USSR. ,1992 Canadiana ,1991 Chemical Engineering Education ,1989 Colour of Metal Compounds Adam Bartecki, John Burgess, Krzysztof Kurzak, 2000-05-01 The Colour of Metal Compounds is devoted to the qualitative and quantitative treatment of colour in inorganic and coordination compounds In order to understand the use of colour as a source of structural and analytical information the book explains in depth the interrelation between colour and structural properties of compounds Trichromatic colorimetry is introduced as a method for the quantitative evaluation of colour Further chapters cover chromaticity and spectroscopy lanthanides colour centres colour in mineralogy pigments

coloured glass and the colour use in teaching Fully revised from the original Polish edition this book is recommended as a supplementary text for undergraduate and graduate level courses on transition metal chemistry coordination chemistry spectroscopy and colour chemistry It will also be of interest to researchers in chemistry physics mineralogy and the pigment and glass industry *Dissertation Abstracts International* ,2003 **Druggists' Circular and Chemical Gazette** ,1881 Includes Red book price list section title varies slightly issued semiannually 1897 1906 **American Druggists' Circular and Chemical Gazette** ,1880 **The United States Catalog** Mary Burnham,Carol Hurd,1928 Physics Briefs ,1993 Russian Journal of Inorganic Chemistry ,1998 Scientific American ,1873 Monthly magazine devoted to topics of general scientific interest Current Index to Journals in Education ,1989 **Russian Journal of Physical Chemistry** ,1991 Energy ,1980 A selection of annotated references to unclassified reports and journal articles that were introduced into the NASA scientific and technical information system and announced in Scientific and technical aerospace reports STAR and International aerospace abstracts IAA **Metals Abstracts Index** ,1984 **Forthcoming Books** Rose Army,1992 The Publishers' Trade List Annual ,1985

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