CHAPTER 14 STUDY GUIDE

Mixtures and Solutions

Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write true or false.

Salter	A solution is a mixture containing particles that settle out of the mixture if left undisturbed.	
Trust	The most abundant substance in a colloid is the dispersion medium.	
tatue	A colloid can be separated by filtration.	
true	 A solid emulsion consists of a liquid dispersed in a solid. 	
Tue	Whipped cream is an example of a foam.	
Nation	6. In an aerosol, the dispersing medium is a liquid.	
Tue	Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.	
Tue	 Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces. 	
floitus	Stirring an electrolyte into a colloid stabilizes the colloid.	
Tive	10. Colloids demonstrate the Tyndall effect.	

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter	157 150 1	~	\rightarrow
12. Between I nm and 1000 nm in diameter	7		
13. More than 1000 nm in diameter		>	
14. Settle out if undisturbed			
15. Pass through standard filter paper			
16. Lower vapor pressure			\
17. Scatter light			

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