

AP Chemistry Semester 1 Exam Study Guide

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. When the following equation is balanced, the coefficient of HNO_3 is _____.



- a. 1
- b. 2
- c. 3
- d. 5
- e. 4

- _____ 2. Predict the product in the combination reaction below.



- a. AlN
- b. Al_3N
- c. AlN_3
- d. Al_3N_2
- e. AlN_3

- _____ 3. What is the empirical formula of a compound that contains 27.0% S, 13.4% O, and 59.6% Cl by mass?

- a. SOCl
- b. SOCl_2
- c. S_2OCl
- d. SO_2Cl
- e. ClSO_4

- _____ 4. A compound is composed of only C, H, and O. The combustion of a 0.519-g sample of the compound yields 1.24 g of CO_2 and 0.255 g of H_2O . What is the empirical formula of the compound?

- a. $\text{C}_6\text{H}_6\text{O}$
- b. $\text{C}_3\text{H}_3\text{O}$
- c. CH_3O
- d. $\text{C}_2\text{H}_5\text{O}_3$
- e. $\text{C}_2\text{H}_5\text{O}_2$

- _____ 5. Propane (C_3H_8) reacts with oxygen in the air to produce carbon dioxide and water. In a particular experiment, 38.0 grams of carbon dioxide are produced from the reaction of 22.05 grams of propane with excess oxygen. What is the % yield in this reaction?

- a. 38.0
- b. 57.6

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Kenneth W. Raymond



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