
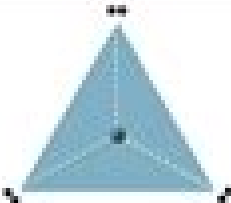
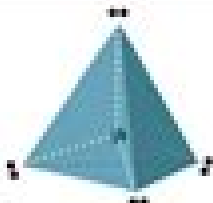
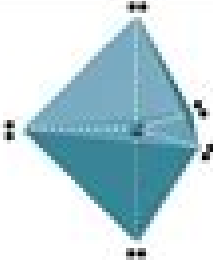




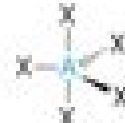
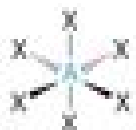



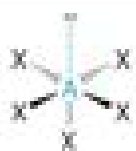


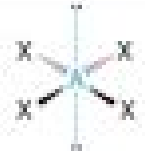



Electron Groups	2	3	4	5	6
Molecular Geometry	 <p>Linear</p>	 <p>Trigonal planar</p>	 <p>Tetrahedral</p>	 <p>Trigonal bipyramidal</p>	 <p>Octahedral</p>
Zero Lone Pairs	 <p>Linear AX_2</p>	 <p>Trigonal planar AX_3</p>	 <p>Tetrahedral AX_4</p>	 <p>Trigonal bipyramidal AX_5</p>	 <p>Octahedral AX_6</p>
One Lone Pair		 <p>Bent (V-shaped) AX_2E</p>	 <p>Trigonal pyramidal AX_3E</p>	 <p>Seesaw AX_5E One axial lone pair</p>	 <p>Square pyramidal AX_6E</p>
Two Lone Pairs		 <p>Bent (V-shaped) AX_2E_2</p>	 <p>T-shaped AX_5E_2 Two axial lone pairs</p>	 <p>Square planar AX_6E_2</p>	
Three Lone Pairs	 <p>Linear AX_3E_3 Three axial lone pairs</p>				

Chem Think Molecular Shapes

**James M. Postma, Julian Roberts, J.
Leland Hollenberg**



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The VSEPR Model of Molecular Geometry Ronald J Gillespie, Istvan Hargittai, 2013-03-21 Valence Shell Electron Pair Repulsion VSEPR theory is a simple technique for predicting the geometry of atomic centers in small molecules and molecular ions This authoritative reference was written by Istvan Hartigai and the developer of VSEPR theory Ronald J Gillespie In addition to its value as a text for courses in molecular geometry and chemistry it constitutes a classic reference for professionals Starting with coverage of the broader aspects of VSEPR this volume narrows its focus to a succinct survey of the methods of structural determination Additional topics include the applications of the VSEPR model and its theoretical basis Helpful data on molecular geometries bond lengths and bond angles appear in tables and other graphics *Molecular Geometry* Alison Rodger, Mark Rodger, 2014-05-16 Molecular Geometry discusses topics relevant to the arrangement of atoms The book is comprised of seven chapters that tackle several areas of molecular geometry Chapter 1 reviews the definition and determination of molecular geometry while Chapter 2 discusses the unified view of stereochemistry and stereochemical changes Chapter 3 covers the geometry of molecules of second row atoms and Chapter 4 deals with the main group elements beyond the second row The book also talks about the complexes of transition metals and f block elements and then covers the organometallic compounds and transition metal clusters The last chapter tackles the consequences of small local variations in geometry The text will be of great use to chemists who primarily deal with the properties of molecules and atoms **Shape in Chemistry** Paul G. Mezey, 1993-08-26 Shape in Chemistry looks at molecular shape from a unique perspective It introduces the reader to the topological concepts and methods of precise shape characterization that are applicable for direct non visual description and analysis of general molecular shapes The author provides a pictorial introduction to all the topological tools necessary for the subjects discussed Mathematical description is also provided at an easily comprehensible level New concepts are introduced beginning at the familiar level of stereochemistry and lead on to more advanced topological shape analysis methods The structure of the book reflects the author's desire to bring the reader to an early appreciation of the power of topology in chemistry After a brief review of the quantum chemical concept the author compares the merits of visual computer graphics methods and nonvisual algorithmic shape analysis methods The book ends with the concepts of approximate symmetry and various generalizations of symmetry Shape in Chemistry is surely destined to become standard reading in the field It presents a valuable addition to the literature on shape and modeling of molecules for non specialists organic physical and medical chemists researchers in various aspects of QSAR and pharmacological drug design and advanced undergraduate and graduate students **The VSEPR Model of Molecular Geometry** Ronald James Gillespie, István Hargittai, 1991-01 **The Electronic Interpretation of Molecular Shapes** B. Deb, 1969 *Models of Molecular Shapes/VSEPR Theory and Orbital Hybridization* James M. Postma, Julian Roberts, J. Leland Hollenberg, 2000-01-15 Molecular Geometry Ronald James Gillespie, 1972

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