

Final Exam Review Chem 101

Each week is broken into segments (a, b and c) based on the different learning sections in each week. Highlighted items are things to memorize, but remember and be able to address problems about all the concepts listed, and be able to apply all the equations listed.

At the end of the table, there are copies of some of the slides from throughout the semester with important information to remember/memorize.

Week	Concepts	Equations/Calculations
2a	Classify matter as element, molecule or compound	
	Classify matter as pure substance (element and compound) or mixture (homogeneous or heterogeneous)	
	Distinguish between physical and chemical properties and changes.	
	Identify unique properties of solids, liquids and gases	
2b	Know distinguishing properties of protons, neutrons and electrons.	
	Know names and symbols of elements on slide #10	
	Determine number of protons, neutrons and electrons from a chemical symbol.	
	Understand difference between atomic number, mass number and atomic mass	
	Calculate atomic mass of an element (see equation below)	
2c	Use periodic table to identify the group number (column) and period (row) of any element.	
	Know the common names of group 1,2,17 and 18 (alkali metal, alkaline earth metal, halogen, noble gas)	
	Identify on periodic table and properties of metals, nonmetals, and metalloids, and gases.	
3a	Describe the difference between a continuous and discontinuous light spectrum	$C = (\text{wavelength}) \times (\text{frequency})$ $C = \text{speed of light, } 3.0 \times 10^8 \text{ m/s}$ $E = (h) \times (\text{frequency})$ $h = \text{Planck's} = 6.626 \times 10^{-34} \text{ m}^2\text{kg/s}$ $E = \text{energy}$
3b	Explain orbital, subshell and shell of electrons	
	Identify an s, p, d and f orbital	
	Write electron configuration for an element.	
3c	Predict the trend in ionization energy and atomic radius of the elements.	
4a	Use valence electrons to find Electron dot structures of elements row 1-3	
	Use octet rule to predict ion formation	
	Write electron configuration of ions	
	Name ions (see list below)	
4b	Write names and formulas for ionic compounds	
4c	Write names and formulas for ionic compounds, including transition metals and polyatomic ions. (see list below)	
5a	Explain difference between ionic and covalent compounds.	
	Write names and formulas for covalent compounds.	
5b	Draw Lewis structure of covalent molecules.	
	Describe electron pair and molecular geometry of a	

Chem 101 Final Exam Review

Daniela Niemeyer



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Chem 101 Final Exam Review Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chem 101 Final Exam Review**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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