

CHEM 1411 - FINAL EXAM

PART I Multiple Choice (2 points each)

Please **DO NOT** write or mark on this paper. Mark your answer on your *Scantron*.

- The value seven hundred fifty million with four significant figures is written as
A. 7500×10^6 B. 7.500×10^8 C. 750×10^6 D. 750,000,000 E. 7500×10^8
- Which of the following is **NOT** an element of the fourth period in the periodic table?
A. Co B. V C. Mg D. Ca E. Kr
- To obtain 5.66×10^{21} atoms of nickel, you would weigh
A. 0.552 g B. 1.81 g C. 106 g D. 9.64×10^{19} g E. 1.04×10^{-20} g.
- How many grams of magnesium contain the same number of atoms as 20.04g of calcium?
A. 12.16 g B. 20.04 g C. 24.30 g D. 40.08 g E. 48.60 g
- Which of the following series represents only known stable metal ions?
A. Fe^{2+} , Fe^{3+} , K^{2+} B. Mg^{2+} , Ba^{3+} , Na^+ C. Li^{2+} , Na^+ , Al^{3+}
D. Fe^{2+} , Sr^{2+} , Mg^{2+} E. Li^{2+} , Ca^{2+} , Al^{3+}
- Which formula represents the binary compound formed by sodium and tellurium?
A. Na_2Te B. NaTe C. Na_3Te D. Na_3Te_2 E. NaTe_2
- What is the correct name of KClO_4 ?
A. potassium chlorate B. potassium perchlorite C. potassium perchlorate
D. potassium hypochlorate E. potassium hypochlorite
- Which of the following compounds is 36.4% oxygen by mass?
A. N_2O B. NO C. N_2O_3 D. N_2O_4 E. N_2O_5
- The molar mass of barium nitrate is
A. 199.21 g/mol. B. 229.32 g/mol. C. 261.35 g/mol. D. 336.61 g/mol. E. 398.62 g/mol.

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RS Peters



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