

STOICHIOMETRY PRACTICE WORKSHEET

1. Using the following equation:



Calculate how many grams of iodine are needed to prepare 28.6 grams of ICl by this reaction.

2. Using the following equation:



How many moles and grams of KMnO_4 are needed for this reaction on 11.4 grams of KNO_2 ?

3. Using the following equation:



How many moles and grams of oxygen (O_2) are needed to react with 56.8 grams of ammonia by this reaction?

4. Using the following equation:



Calculate the number of moles and the number of grams of iodine (I_2) that can be made this way from 16.4 grams of NaIO_3 .

Chem 20 Sample Stoichiometry Questions

Lawrie Ryan, Roger Norris



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Rapid Review of Chemistry for the Life Sciences and Engineering Armen S. Casparian, Gergely Sirokman, Ann Omollo, 2021-12-07 Designed to demystify chemistry for the non chemist Rapid Review of Chemistry for the Life Sciences and Engineering is a useful reference manual for life scientists and engineers who may have forgotten a formula principle or concept in the college chemistry taken a few years ago With over 100 solved examples from balancing chemical reactions doing stoichiometry and understanding nomenclature rules in both organic and inorganic chemistry to calculating half lives in kinetics or radioactive decay schemes understanding colligative properties of solutions and interpreting toxicities of hazardous materials this book is intended to make reviewing and understanding chemistry much clearer and easier Relevant diagrams are in color and solved examples are organized by subject topic and cross referenced by page and chapter number It may also serve as a concise go to sidekick for students who are not chemistry majors taking chemistry at the college level and having difficulty understanding the scope focus language or equations in their chemistry textbook Armed with select contemporary applications it is written in the hope to bridge a gap between chemists and non chemists so that they may communicate with and understand each other Chapters 1 10 are designed to contain the standard material in an introductory college chemistry course Chapters 11 15 present applications of chemistry that should interest and appeal to scientists and engineers engaged in a variety of fields Additional features More than 100 solved examples clearly illustrated and explained with SI units and conversion to other units using conversion tables included Assists the reader to understand organic and inorganic compounds along with their structures including isomers enantiomers and congeners of organic compounds Provides a quick and easy access to basic chemical concepts and specific examples of solved problems Ideal sidekick for students who are non chemistry majors taking intro college chemistry needing clear concise explanations This concise user friendly review of general and organic chemistry with environmental applications will be of interest to all disciplines and backgrounds

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