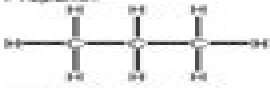
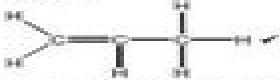


Question	Answers	Notes	Total
5. a	Propane:  AND Propene: 		1
5. b. i	$C_3H_6 + Br_2 \rightarrow C_3H_6Br_2 + HBr$ ✓ «sunlight/UV» OR high temperature ✓	Do not accept 'reflux' for M2.	2
5. b. ii	$C_3H_6 + Br_2 \rightarrow C_3H_5Br_2$ ✓		1
5. b. iii	Propane: «free radical» substitution / S_N AND Propene: «electrophilic» addition / A_E ✓	Award mark even if incorrect type of substitution/ addition given.	1

1. d	acidified «potassium» dichromate(VI) / H^+ AND $K_2Cr_2O_7$ / H^+ AND $Cr_2O_7^{2-}$ OR acidified potassium «manganate(VII)» / H^+ AND $KMnO_4$ / H^+ AND MnO_4^- ✓	Accept H_2SO_4 or H_3PO_4 for H^+ . Accept 'permanganate' for 'manganate(VII)'.	1
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4. a	I: carbonate ✓ II: phenyl ✓ III: carboxyl / carboxy ✓ IV: hydroxyl ✓	Award [2] for all four correct, [1] for two or three correct. Do not allow 'benzene'. Do not allow 'carboxylic/alkanoic acid'. Do not allow 'alcohol' or 'hydroxide'.	2 max.
b. iii	$(0.5)(40 - 20 - 2) = 9$ ✓		1
b. iv	A: C-H and B: C=O		1
b. v	O-H and N-H ✓ Frequencies/stretches due to O-H and N-H occur above 3200 cm^{-1} which are not present in B; of base ✓		2
c. i	1:1:6 ✓		1

Question	Answers	Notes	Total
c. iii			1
c. iii	$CH_3OCH_2CH_3$ ✓		1
c. iv	Similarity: both have fragment corresponding to $(M_r - 15)^+$ / both have $m/z = 45$ ✓ Difference: X has fragment corresponding to $(M_r - 17)^+$ / X has $m/z = 43$ OR X has fragment corresponding to $(M_r - 43)^+$ / X has $m/z = 17$ OR Y has fragment corresponding to $(M_r - 31)^+$ / Y has $m/z = 29$ OR Y has fragment corresponding to $(M_r - 29)^+$ / Y has $m/z = 31$ ✓	Allow both have same molecular ion peak (M^+) / both have $m/z = 60$. However in practice the molecular ion peak is of low abundance and difficult to observe for propan-2-ol.	2



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