

Chemistry Study Guide

The horizontal rows are called periods. 7 periods

The vertical columns are called groups. 18 groups

118 elements

The periodic table is arranged according to increasing atomic numbers.

There are 8 main groups in the periodic table:

Alkali metals:

- group 1
- very reactive
- silvery and soft
- nonmetals
- not found free in nature

Alkaline earth metals:

- group 2
- less reactive
- hard and dense
- high melting points
- not found free in nature

Transition metals:

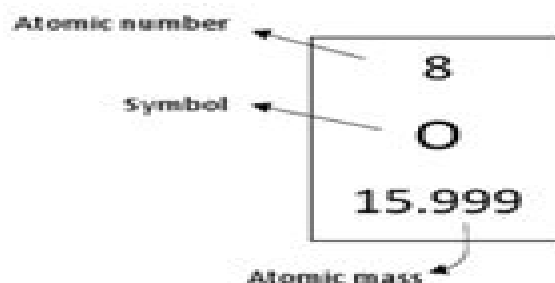
- groups 3 to 12 (d-block)
- good conductors of electricity
- high luster
- less reactive
- can be found free in nature

Halogens:

- group 17
- most reactive of the nonmetals
- react strongly with metals to produce salts

Noble gases:

- group 8
- relatively unreactive because they have a full outer shell
- very low electronegativity



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Chem Study Guide:

Study Guide for Chemical Principles [by] Steven S. Zumdahl Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples

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