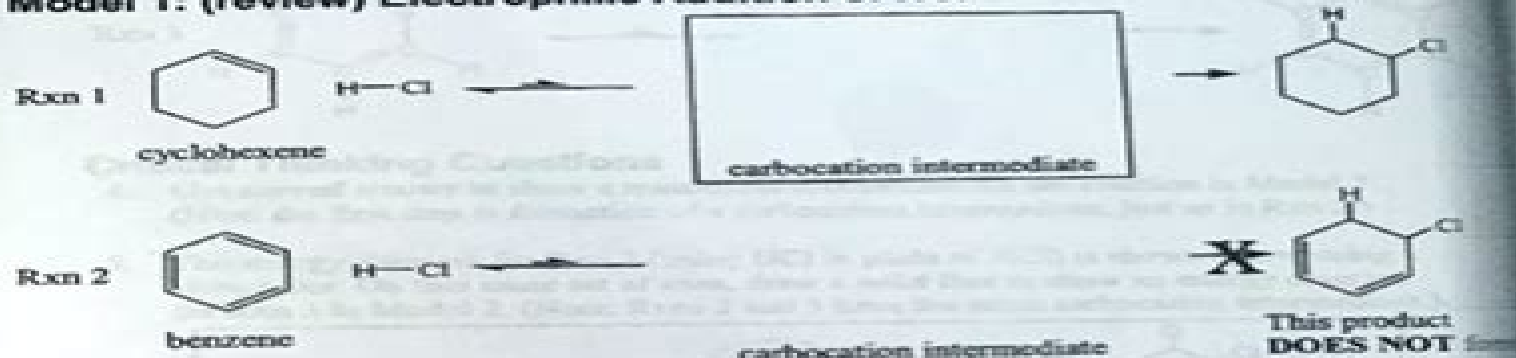


ChemActivity 29

Part A: Electrophilic Aromatic Substitution

(What products are formed when a strong electrophile is added to benzene?)

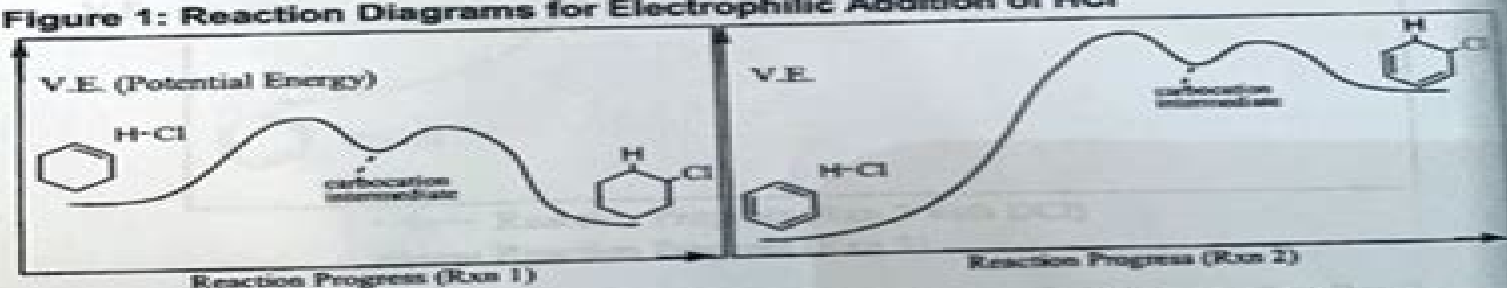
Model 1: (review) Electrophilic Addition of HCl



Critical Thinking Questions

- For Rxn 1 (above) draw curved arrows showing the mechanism of electrophilic addition of HCl. Include an appropriate carbocation intermediate in the box above.

Figure 1: Reaction Diagrams for Electrophilic Addition of HCl



- Rxn 1 is slightly down-hill in terms of energy. Rxn 2 is very up-hill in terms of energy (see Figure 1). Construct an explanation for the large difference in energy between the reactants and the product in Rxn 2.
- Draw the carbocation that would form in Rxn 2. Explain why this carbocation goes back to the starting material (H-Cl and benzene) instead of forming the product.

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[Anonymus AC00033125]



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1. Understanding the eBook Chemactivity 29 Key
 - The Rise of Digital Reading Chemactivity 29 Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemactivity 29 Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemactivity 29 Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemactivity 29 Key
 - Personalized Recommendations
 - Chemactivity 29 Key User Reviews and Ratings
 - Chemactivity 29 Key and Bestseller Lists
5. Accessing Chemactivity 29 Key Free and Paid eBooks
 - Chemactivity 29 Key Public Domain eBooks
 - Chemactivity 29 Key eBook Subscription Services
 - Chemactivity 29 Key Budget-Friendly Options
6. Navigating Chemactivity 29 Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemactivity 29 Key Compatibility with Devices
 - Chemactivity 29 Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemactivity 29 Key
 - Highlighting and Note-Taking Chemactivity 29 Key
 - Interactive Elements Chemactivity 29 Key
8. Staying Engaged with Chemactivity 29 Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chemactivity 29 Key
- 9. Balancing eBooks and Physical Books Chemactivity 29 Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemactivity 29 Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemactivity 29 Key
 - Setting Reading Goals Chemactivity 29 Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemactivity 29 Key
 - Fact-Checking eBook Content of Chemactivity 29 Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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