2. Chemistry

Core Topics:

- Some Basic Concepts of Chemistry: Mole concept, stoichiometry, atomic mass, empirical formulas.
- Structure of Atom: Bohr's model, quantum numbers, electronic configuration.
- Periodic Classification: Modern periodic table, trends in atomic and ionic radii, ionization energy.
- Chemical Bonding: Types of bonds, VSEPR theory, hybridization.
- Thermodynamics: Enthalpy, entropy, Gibbs free energy.

Key Formulae:

- PV=nRTPV = nRTPV=nRT (Ideal gas law)
- Enthalpy change: ΔH=ΔU+PΔV\Delta H = \Delta U + P\Delta VΔH=ΔU+PΔV

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