complete look equation: $HCI + NOOH \rightarrow NoCI + HaO$ $Complete look equation: H^+ + CI^- + NoC^- + OH^ closs + CI^- + NI$
complete took equation: H+ + CV + NoV + OH clos + (1 + 1)
net innic equation H + + OH - > H > 0
The state of the s
someral reaction type: Double replacement/Displacement
YI. Heating Cu with Atmospheric O.
appearance the unreacted Cu mesh: Shippy Copply
appearance of the Cu mesh after heating and cooling: AULI dowk brown
chemical equation: CV+02 -> 2CVO
personal reaction type: Combico tion
Which species are oxidized?C.v
VII. Reacting CuSO ₄ Solution with Steel Wool (Fe)
appearance of the steel wool (Fe) before reaction:
appearance of the CuSO, solution before reaction: Light blue
appearance of the steel wool (Fe) after the reaction has occurred
ppearance of the CuSO, solution after the reaction has occurred: Light blue liquid
olecular equation: CUSOH+Fe -> CV+Fe504
oplete ionic equation: CV+2 (ng) + 50 = (ng) + Fe (s) -> CV (s) + Fe (sous (ng
onic equation: Cut2 (sig) + Fe(s) -> Cu(s) + Fe+2 (sig)
ator ions: 50 = 100) general reaction type: Single Displacement
species are oxidized? Fe Which species are reduced? Cu+2

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NARAYAN CHANGDER

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