

Naming Ionic Compounds Worksheet

Name the following ionic compounds:

- 1) NH_4Cl _____
- 2) $\text{Fe}(\text{NO}_3)_3$ _____
- 3) TiBr_3 _____
- 4) Cu_3P _____
- 5) SnSe_2 _____
- 6) GaAs _____
- 7) $\text{Pb}(\text{SO}_4)_2$ _____
- 8) $\text{Be}(\text{HCO}_3)_2$ _____
- 9) $\text{Mn}_2(\text{SO}_4)_3$ _____
- 10) $\text{Al}(\text{CN})_3$ _____

Write the formulas for the following compounds:

- 11) chromium (VI) phosphate _____
- 12) vanadium (IV) carbonate _____
- 13) tin (II) nitrite _____
- 14) cobalt (III) oxide _____
- 15) titanium (II) acetate _____
- 16) vanadium (V) sulfide _____
- 17) chromium (III) hydroxide _____
- 18) lithium iodide _____
- 19) lead (II) nitride _____
- 20) silver bromide _____

Lots of Ionic Naming Practice Problems

Name the following ionic compounds:

- 1) NaBr _____
- 2) $\text{Sc}(\text{OH})_3$ _____
- 3) $\text{V}_2(\text{SO}_4)_3$ _____
- 4) NH_4F _____
- 5) CaCO_3 _____
- 6) NiPO_4 _____
- 7) Li_2SO_3 _____
- 8) Zn_3P_2 _____
- 9) $\text{Sn}(\text{C}_6\text{H}_5\text{O}_2)_2$ _____
- 10) Cu_2O _____
- 11) Ag_3PO_4 _____
- 12) YClO_3 _____
- 13) SnS_2 _____
- 14) $\text{Ti}(\text{CN})_4$ _____
- 15) KMnO_4 _____
- 16) Pb_2N_2 _____
- 17) CoCO_3 _____
- 18) CdSO_3 _____
- 19) $\text{Cu}(\text{NO}_2)_2$ _____
- 20) $\text{Fe}(\text{HCO}_3)_2$ _____

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