

CHAPTER 14 STUDY GUIDE

Mixtures and Solutions

Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write *true* or *false*.

- false 1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed.
- true 2. The most abundant substance in a colloid is the dispersion medium.
- false 3. A colloid can be separated by filtration.
- true 4. A solid emulsion consists of a liquid dispersed in a solid.
- true 5. Whipped cream is an example of a foam.
- false 6. In an aerosol, the dispersing medium is a liquid.
- true 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.
- true 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces.
- false 9. Stirring an electrolyte into a colloid stabilizes the colloid.
- true 10. Colloids demonstrate the Tyndall effect.

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter		<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>
12. Between 1 nm and 1000 nm in diameter	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>
13. More than 1000 nm in diameter	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>	
14. Settle out if undisturbed	<input checked="" type="checkbox"/>		
15. Pass through standard filter paper	<input checked="" type="checkbox"/>		
16. Lower vapor pressure			<input checked="" type="checkbox"/>
17. Scatter light	<input checked="" type="checkbox"/>		

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Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise

Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text Study Guide and Selected Solutions Manual for Basic Chemistry Karen

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