

## PART 2: TYPES OF BONDS

A chemical bond is an attraction between two or more atoms. Every type of chemical bond is based on the electronegativity of the atoms involved in the bonds. Types of bonds/interactions include:

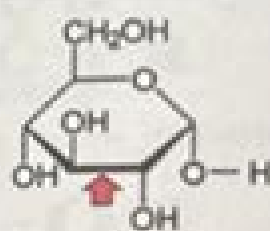
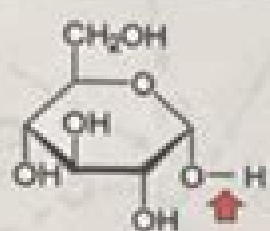
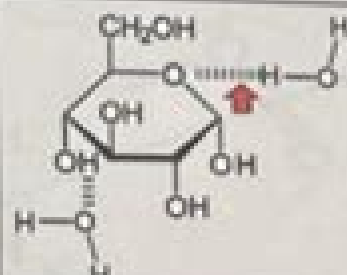

**Ionic Bonds:** form when one atom steals an electron from another, resulting in two oppositely charged ions. The ions are electrostatically attracted to each other (opposites attract).

**Covalent bonds:** form when two atoms share electrons.

**Hydrogen bonds:** are a type of electrostatic attraction between oppositely charged dipoles. Dipoles are not full charges.

1. Fill in the bond names below:

Bond Name:

2. The hydrogen bond example above is drawn with a dashed line. Hydrogen bonds should always be demonstrated with a dashed line. Draw three water molecules below and the hydrogen bonds occurring between them.

# Chemical Bonds Packet Section 2 Types Of Bonds Answers

**Bowker Editorial Staff**



## **Chemical Bonds Packet Section 2 Types Of Bonds Answers:**

**Chapter Resource 2 Chemistry of Life Biology** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004

*Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science*, 2003-11  
Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **American Druggists' Circular and Chemical Gazette**, 1922

**Discover Science: Teacher's annotated edition**, 1991 Science content helps develop the skills needed to understand how science works learn new concepts solve problems and make decisions in today's technological society **The Software Encyclopedia 2000** Bowker Editorial Staff, 2000-05 *10 in One Study Package for CBSE Science Class 10 with Objective Questions & 3 Sample Papers 4th Edition* Disha Experts, 2020-08-08 **10 in One Study Package for CBSE Science Class 10 with Objective Questions & 3 Sample Papers 3rd Edition** Disha Experts, 2019-05-16 As per the latest CBSE Notification Class 10 Science Board Exams will feature MCQs Assertion Reasoning Qns in the 20 Qns of the 1 Mark category The 3rd edition of the book 10 in ONE CBSE Study Package Mathematics class 10 with 3 Sample Papers has 10 key ingredients that will help you achieve success 1 Chapter Utility Score CUS 2 Exhaustive Theory with Concept Maps 3 Text Book exercises 4 VSA SA LA Questions 5 Past year questions including 2017 2018 Solved papers 6 HOTS Value based Exemplar 7 Past NTSE Exemplar MCQs as required with the latest change in CBSE pattern 8 Objective Questions VSA MCQs Assertion Reasoning etc 9 Important Formulas Terms Definitions 10 Latest Pattern 2019 20 3 Sample Papers with detailed solutions **The Package Engineering Handbook** Walter Stern, 1954 *Glencoe Science* McGraw-Hill Staff, 2001-06

Notices of Judgment Under the Food and Drugs Act United States. Food and Drug Administration, 1923 Apothecary and New England Druggist, 1909 Basic Chemistry Steven S Zumdahl, 1996 *Introductory Chemistry* Steven S. Zumdahl, 1996 Drug & Chemical Markets, 1916 Drug and Chemical Markets, 1916 **Exploring Earth and Space** Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter **Traffic World and Traffic Bulletin**, 1928

**Operation & Maintenance**, 1922 **Army and Navy Gazette**, 1877 The Traffic World, 1928

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