

Chemical Bonding Worksheet

Ionic Bond
Covalent Bond
Metallic Bond

between a Metal and Non-Metal
between a Non-Metal and Non-Metal
between a Metal and Metal

Compound	Name	Oxidation states of each atom	Bond Type
NO ₂	nitrogen dioxide	N = +4 O = -2	covalent
NaCl	sodium chloride	Na = +1 Cl = -1	ionic
SO ₂	sulfur dioxide	S = +4 O = -2	covalent
PO ₄ ³⁻	phosphate anion	P = +5 O = -2	covalent
MgBr ₂	magnesium bromide	Mg = +2 Br = -1	ionic
CaO	calcium oxide	Ca = +2 O = -2	ionic
H ₂ O	water	H = +1 O = -2	covalent
K ₂ O	potassium oxide	K = +1 O = -2	ionic
Cu-Zn alloy	brass	Cu = 0 Zn = 0	metallic
O ₂	oxygen gas	O = 0	covalent
CuCl ₂	copper(II) chloride	Cu = +2 Cl = -1	ionic
NO ₂ ⁻	nitrite anion	N = +3 O = -2	covalent
TiO ₂	titanium (IV) oxide	Ti = +4 O = -2	ionic
HF	hydrofluoric acid	H = +1 F = -1	covalent
Rb ₂ S	rubidium sulfide	Rb = +1 S = -2	ionic
Au-Pt mixture	white gold	Au = 0 Pt = 0	metallic
Fe ₂ O ₃	iron (III) oxide	Fe = +3 O = -2	ionic
C ₆ H ₁₂ O ₆	glucose	C = +4 H = -1 O = -2	covalent

Chemical Bonding Review Answer Key

The Princeton Review



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Table of Contents Chemical Bonding Review Answer Key

1. Understanding the eBook Chemical Bonding Review Answer Key
 - The Rise of Digital Reading Chemical Bonding Review Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Bonding Review Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Bonding Review Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Bonding Review Answer Key
 - Personalized Recommendations
 - Chemical Bonding Review Answer Key User Reviews and Ratings
 - Chemical Bonding Review Answer Key and Bestseller Lists
5. Accessing Chemical Bonding Review Answer Key Free and Paid eBooks
 - Chemical Bonding Review Answer Key Public Domain eBooks
 - Chemical Bonding Review Answer Key eBook Subscription Services
 - Chemical Bonding Review Answer Key Budget-Friendly Options
6. Navigating Chemical Bonding Review Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Bonding Review Answer Key Compatibility with Devices
 - Chemical Bonding Review Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Bonding Review Answer Key
 - Highlighting and Note-Taking Chemical Bonding Review Answer Key
 - Interactive Elements Chemical Bonding Review Answer Key
8. Staying Engaged with Chemical Bonding Review Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chemical Bonding Review Answer Key
- 9. Balancing eBooks and Physical Books Chemical Bonding Review Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Bonding Review Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemical Bonding Review Answer Key
 - Setting Reading Goals Chemical Bonding Review Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemical Bonding Review Answer Key
 - Fact-Checking eBook Content of Chemical Bonding Review Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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