



Reinforcement

Chemical Changes

Read the passage and answer the following questions:



1. What are the products of this chemical reaction?

As mentioned in the introduction, the burning of methane in this chemical reaction, the products are water and carbon dioxide. The reaction is exothermic, meaning it releases heat. The reaction is also a combustion reaction, meaning it involves the burning of a substance.

2. What is the reactant in this chemical reaction? Is it a solid, liquid, gas, or plasma? Is it a pure substance or a mixture?

3. What is the physical state of the reactant and the product?

Hydrogen peroxide is a colorless liquid. It is a pure substance. The products, water and carbon dioxide, are both gases. They are also pure substances. The reaction is exothermic, meaning it releases heat. The reaction is also a combustion reaction, meaning it involves the burning of a substance.

4. According to the law of conservation of mass, if 100g of hydrogen peroxide decomposes, how much water and carbon dioxide will be produced?

100 grams



5. What is the balanced chemical equation for this reaction? Is it?

6. What is the balanced chemical equation for this reaction? Is it?

7. What are the physical states of the reactant and the product?

Hydrogen peroxide is a liquid.
Water and oxygen are gases.

8. What is the physical state of the reactant and the product?

9. What is the balanced chemical equation for this reaction? Is it?

Hydrogen peroxide is a liquid. Water and oxygen are gases.

10. According to the law of conservation of mass, if 100g of hydrogen peroxide decomposes, how much water and carbon dioxide will be produced?

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