

CHEMISTRY 11 – WORD EQUATIONS AND TYPES OF CHEMICAL REACTIONS WORKSHEET

- 1) Translate the following word equations into chemical symbols and balance the resulting equations. Do not include the phases.

(a) Potassium + water \rightarrow potassium hydroxide + hydrogen



(b) Nitrogen + hydrogen \rightarrow ammonia



- 2) In each of the following write a balanced chemical equation, including the phases.

(a) Liquid phosphoric acid reacts with aqueous barium hydroxide to give water and a precipitate of barium phosphate



(b) Solid aluminium oxide and aqueous sulphuric acid produce water and aqueous aluminium sulphate



(c) Nitrogen trifluoride gas and hydrogen gas react to form nitrogen gas and gaseous hydrogen fluoride



(d) Powdered sodium carbonate and aqueous hydrobromic acid react to form carbon dioxide gas, aqueous sodium bromide and water



- 3) Classify the following reactions.

Reaction	Type
$2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$	Combustion
$\text{Mg} + \text{CuSO}_4 \rightarrow \text{MgSO}_4 + \text{Cu}$	Single Replacement
$4\text{Na} + \text{O}_2 \rightarrow 2\text{Na}_2\text{O}$	Synthesis
$\text{Sn}(\text{OH})_4 + 4\text{HBr} \rightarrow 4\text{H}_2\text{O} + \text{SnBr}_4$	Neutralization
$2\text{Fe}(\text{NO}_3)_3 + 3\text{MgS} \rightarrow \text{Fe}_2\text{S}_3 + 3\text{Mg}(\text{NO}_3)_2$	Double Replacement
$2\text{N}_2\text{O} \rightarrow 2\text{N}_2 + \text{O}_2$	Decomposition

Chemical Reactions And Equations Answers Prentice Hall

Richard C. Aiken



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Pao, 2012-12-06 In response to the growing use of reaction diffusion problems in many fields this monograph gives a systematic treatment of a class of nonlinear parabolic and elliptic differential equations and their applications these problems It is an important reference for mathematicians and engineers as well as a practical text for graduate students

Elementary Feedback Stabilization of the Linear Reaction-Convection-Diffusion Equation and the Wave Equation Weijiu Liu, 2009-12-01 Unlike abstract approaches to advanced control theory this volume presents key concepts through concrete examples Once the basic fundamentals are established readers can apply them to solve other control problems of partial differential equations **The Method of Weighted Residuals and Variational Principles** Bruce A. Finlayson, 2013-12-30 This classic book covers the solution of differential equations in science and engineering in such a way as to provide an introduction for novices before progressing toward increasingly more difficult problems The Method of Weighted Residuals and Variational Principles describes variational principles including how to find them and how to use them to construct error bounds and create stationary principles The book also illustrates how to use simple methods to find approximate solutions shows how to use the finite element method for more complex problems and provides detailed information on error bounds Problem sets make this book ideal for self study or as a course text *Introduction to Ordinary Differential Equations* Roger McCann, 1982 [Resources for Teaching Middle School Science](#) Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also

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Reaction-diffusion Equations and Their Applications to Biology N. F. Britton, 1986 Although the book is largely self contained some knowledge of the mathematics of differential equations is necessary Thus the book is intended for mathematicians who are interested in the application of their subject to the biological sciences and for biologists with some mathematical training It is also suitable for postgraduate mathematics students and for undergraduate mathematicians taking a course in mathematical biology Increasing use of mathematics in developmental biology ecology physiology and many other areas in the biological sciences has produced a need for a complete mathematical reference for laboratory practice In this volume biological scientists will find a rich resource of interesting applications and illustrations of various mathematical techniques that can be used to analyze reaction diffusion systems Concepts covered here include systems of ordinary differential equations conservative systems the scalar reaction diffusion equation analytic techniques for systems of parabolic partial differential equations bifurcation theory asymptotic methods for oscillatory systems singular perturbations macromolecular carriers asymptotic techniques

Elementary Differential Equations Werner E. Kohler, Lee W. Johnson, 2003 Elementary Differential Equations with Boundary Value Problems integrates the underlying theory the solution procedures and the numerical computational aspects of differential equations in a seamless way that provides students with the necessary framework to understand and solve differential equations Theory is presented as simply as possible with an emphasis on how to use it With an emphasis on linear equations linear and nonlinear equations first order and higher order are treated in separate chapters In developing mathematical models this text guides the student carefully through the underlying physical principles leading to the relevant

mathematics Asking students to use common sense intuition and back of the envelope checks as well as challenging them to anticipate and interpret the physical content of the solution encourage critical thinking MARKET Intended for use in introductory course in differential equations Physical Chemistry John S. Winn,1995 This text presents physical chemistry as a coherent whole rather than a set of disjointed topics and shows how the subject relates to the rest of chemistry and physics It emphasizes physical models as well as mathematical techniques along with both rigorous and approximate order of magnitude problem solving Designed to progress beyond a numerical answer problems expose the physical significance of the situation and teach students how to pose a problem in the first place In addition modern molecular concepts currently unanswered problems in research experimental techniques and new directions in the field are introduced wherever appropriate An orderly progression of thermodynamics carefully builds students knowledge without covering too much too early on Chemical reaction thermodynamics is covered in Chapter 7 after the culmination of thermodynamics with advanced material in Chapter 10 **Holt General Science** William L. Ramsey,1988 Quantitative Sociodynamics Dirk Helbing,2010-11-15 When I wrote the book Quantitative Sociodynamics it was an early attempt to make methods from statistical physics and complex systems theory fruitful for the modeling and understanding of social phenomena Unfortunately the rst edition appeared at a quite prohibitive price This was one reason to make these chapters available again by a new edition The other reason is that in the meantime many of the methods discussed in this book are more and more used in a variety of different elds Among the ideas worked out in this book are 1 a statistical theory of binary social interactions a mathematical formulation of social eld theory which is the basis of social 2 force models a microscopic foundation of evolutionary game theory based on what is known today as proportional imitation rule a stochastic treatment of interactions in evolutionary game theory and a model for the self organization of behavioral 3 conventions in a coordination game It therefore appeared reasonable to make this book available again but at a more affordable price To keep its original character the translation of this book which 1 D Helbing Interrelations between stochastic equations for systems with pair interactions Ph icaA 181 29 52 1992 D Helbing Boltzmann like and Boltzmann Fokker Planck equations as a foundation of behavioral models PhysicaA 196 546 573 1993 2 D Helbing Boltzmann like and Boltzmann Fokker Planck equations as a foundation of beh ioral models PhysicaA 196 546 573 1993 D **Encyclopedia of Optimization** Christodoulos A. Floudas,Panos M. Pardalos,2008-09-04 The goal of the Encyclopedia of Optimization is to introduce the reader to a complete set of topics that show the spectrum of research the richness of ideas and the breadth of applications that has come from this field The second edition builds on the success of the former edition with more than 150 completely new entries designed to ensure that the reference addresses recent areas where optimization theories and techniques have advanced Particularly heavy attention resulted in health science and transportation with entries such as Algorithms for Genomics Optimization and Radiotherapy Treatment Design and Crew Scheduling *The Mathematical Theory of Diffusion and Reaction in Permeable*

Catalysts: Questions of uniqueness, stability, and transient behaviour Rutherford Aris, 1975 **Transport with Chemical Reactions** Pieter Stroeve, William J. Ward, 1981 **The Mathematical Theory of Diffusion and Reaction in Permeable Catalysts: The theory of the steady state** Rutherford Aris, 1975 *Physical Pharmacy* Alfred N. Martin, James Swarbrick, Arthur Cammarata, 1983

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General Chemistry Ralph H. Petrucci, 1972

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