Mole Conversions Practice Answer Key

Molar Mass - Calculate the Molar Mass for the following.

What is the molar mass of Magnesium metal?

24.305 gmal

Calculate the molar mass of NaCl

Nq:20.990 x1 = 20.990 C1:35.453 x1 = 35.45 3

3. What is the molar mass of NaOH?

Na: 1 x 22.990 = 22.990 D: 1 x 15.999 = 15.999 H: 1 x 1.008 = 1.008

Conversions - Calculate the following the conversions. Use your Mole Map to help you!

Determine the number of atoms in 0.50 moles He.

= 13.01 x 10 23 atoms He

Determine the number of moles in 80.00 g O₂

0 2 x 15 999 = 31 998 201

=> 1 mol = 31.9980

Determine the volume of 0.24 mol of NO₂.

4. Determine the number of moles in 7.5 x 1019 formula units of KNOs

75×1019 f.u. KNO, 6.03 ×1023 f.u KNO

Determine the number of moles in 202,22a KNO₃

K 1x 39.098 = 39.098 0: 3x 15.999 = 47.997) 101.102 g/mol N 17 14.007 = 14.007

6. Determine the number of moles in 32.5L of N₂ at STP In a = 101 102 g

Determine the number of moles in 3.4 x 10²⁶ formula units of NaCl.

34 x 10 F. U. Nacl -6.02 ×1023 f.4 Nacl

Determine the moles in 36.5L of carbon monoxide

36.5 L CO x | Imel CO

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